

Gas Stoichiometry Practice Sheet

- 1) For the reaction $2\text{H}_{2(g)} + \text{O}_{2(g)} \rightarrow 2\text{H}_2\text{O}_{(g)}$, how many liters of water can be made from 5 L of oxygen gas and an excess of hydrogen?
- 2) How many liters of water can be made from 55 grams of oxygen gas and an excess of hydrogen at STP?
- 3) How many liters of water can be made from 55 grams of oxygen gas and an excess of hydrogen at a pressure of 12.4 atm and a temperature of 32°C?
- 4) How many liters of water can be made from 34 grams of oxygen gas and 6.0 grams of hydrogen gas at STP? What is the limiting reactant for this reaction?

Gas Stoichiometry Practice Sheet Answers

Pamela Mills, Amina El-Ashmawy

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