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HL Paper 2

Benzoic acid, CeHeCOOH, is another derivative of benzene.

 a. Identify the wavenumber of one peak in the IR spectrum of benzoic acid, using section 26 of the data booklet. 	(1)
b. Identify the spectroscopic technique that is used to measure the bond lengths in solid benzoic acid.	613
 Outline one piece of physical evidence for the structure of the benzene ring. 	[1]
d. Draw the structure of the conjugate base of benzoic acid showing all the atoms and all the bonds.	(2)
e. Outline why both C to O bonds in the conjugate base are the same length and suggest a value for them. Use section 10 of the data booklet	t [2]
\$(i). The pit of an aqueous solution of benzoic acid at 298 K is 2.95. Determine the concentration of hydroxide ions in the solution, using section	n 2 of [2]
the data booklet.	
f(ii)Formulate the equation for the complete combustion of benzoic acid in oxygen using only integer coefficients.	[2]
g. The combustion reaction in (f)(ii) can also be classed as redox. Identify the atom that is oxidized and the atom that is reduced.	(4)
h. Suggest how benzoic acid, M, = 122.13, forms an apparent dimer, M, = 244.26, when dissolved in a non-polar solvent such as hexane.	[1]
 State the reagent used to convert benzoic acid to phenylmethanol (benzyl alcohol), C₆H₃CH₂OH. 	(1)

A student performs a titration to determine the concentration of ethanoic acid, CH₃ COOH, in vinegar using potassium hydroxide.

The pH curve for the reaction is given.



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M Woodhall



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