#### IB Chemistry - SL

#### **Topic 2 Questions**

Consider the composition of the species W, X, Y and Z below. Which species is an anion?

Species	Number of protons	Number of neutrons	Number of electrons
W	9	10	10
×	11	12	11
Y	12	12	12
Z	13	14	10

- A. W
- B. X.
- C. Y
- $D_{ij} = Z_i$

(Total I mark)

- 2. Energy levels for an electron in a hydrogen atom are
  - A. evenly spaced.
  - farther apart near the nucleus.
  - C. closer together near the nucleus.
  - D. arranged randomly.

(Total I mark):

- 3. Which is related to the number of electrons in the outer main energy level of the elements from the alkali metals to the halogens?
  - Group number
  - II. Period number
  - A. Lonly
  - B. Honly
  - C. Both I and II
  - D. Neither I nor II

(Total I mark)

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**Clemens Wendtner** 

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