

ঢাকা বোর্ড

- ১৪। ব্রীকবার্ট বিকারক ব্যবহার করে কোনটি থেকে 2° অ্যালকোহল তৈরি করা যায়?
 (ক) CH_3OH (খ) HCHO
 (গ) $\text{CH}_3\text{CH}_2\text{OH}$ (ঘ) CH_3CHO

- ১৫। অ্যামোনিয়া লোহাটিকে কোনটি অধিক ক্যাথডিক বিভ্রাতি দেওয়া?
 (ক) H_2 (খ) NH_3
 (গ) Hc (ঘ) CO_2

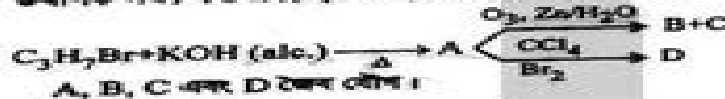
- ১৬। OH^- এর অনুবর্তী আয় কোনটি?
 (ক) O^{2-} (খ) H_2O
 (গ) H_3O^+ (ঘ) O_2

- ১৭। কোন বিদ্যুৎ তড়িৎনের আংশিক তাপ সূত্র মেনে চলে?
 (ক) NH_3 , HCl (খ) C_2H_6 , N_2
 (গ) SO_2 , H_2S (ঘ) SO_2 , O_2

- ১৮। SnO_2 এর আরকত্ব কত?
 (ক) 2
 (খ) 4
 (গ) 6
 (ঘ) 8

- ১৯। পলিত অ্যালুমিনার দ্রব দিতে 30 অ্যাম্পিয়ার বিদ্যুৎ 90 মিনিট ব্যবহৃত করলে কারখোতে কত গ্রাম তামা জমা হয়?
 (ক) 7.27
 (খ) 15.10
 (গ) 21.82
 (ঘ) 45.32

উদ্ভীষ্টক লক্ষ্য কর এবং ২০ ও ২১ নং প্রশ্নের উত্তর দাও :



- ২০। যৌগ 'A' কোনটি?
 (ক) প্রোপানল
 (খ) প্রোপান্যাল
 (গ) প্রোপিন
 (ঘ) প্রোপাইন

- ২১। যৌগ 'B' ও 'C' এর ক্ষেত্রে প্রযোজ্য একা উত্তরই—
 I. ফেহলিং প্রবাদের সাথে বিক্রিয়া করে
 II. 2, 4-DNPH এর সাথে বিক্রিয়া করে
 III. LiAlH_4 দ্বারা 1° অ্যালকোহল তৈরি করে

নিচের কোনটি সঠিক?

- (ক) I
 (খ) I ও II
 (গ) I ও III
 (ঘ) I, II ও III

- ২২। $0.5 \text{ Mol L}^{-1} \text{ H}_2\text{SO}_4$ দ্রবনে H^+ এর ঘনমাত্রা কত নির্দিষ্ট?
 (ক) 10,000
 (খ) 1,000
 (গ) 100
 (ঘ) 10

- ২৩। অ্যামোনিয়াম সোরাইট ও ক্যালসিয়াম অক্সাইড বিক্রিয়া করে STP তে 44-SL NH_3 গ্যাস প্রস্তুত করতে ব্যবহৃত ক্যালসিয়াম অক্সাইডের পরিমাণ কত?
 (ক) 56g
 (খ) 28g
 (গ) 14g
 (ঘ) 7g

- ২৪। কোনটি জারন-বিজারন বিক্রিয়া?
 (ক) $\text{CaCO}_3 \xrightarrow{\Delta} \text{CaO} + \text{CO}_2$
 (খ) $\text{NaOH} + \text{HCl} \longrightarrow \text{NaCl} + \text{H}_2\text{O}$
 (গ) $\text{CaF}_2 + 2\text{AgNO}_3 \longrightarrow 2\text{AgF} + \text{Ca(NO}_3)_2$
 (ঘ) $2\text{H}_2\text{S} + \text{SO}_2 \longrightarrow 2\text{H}_2\text{O} + 3\text{S}$

- ২৫। $\text{Pt} | \frac{1}{2} \text{H}_2(\text{g}) | \text{H}^+(\text{aq}) || \text{Cu}^{2+}(\text{aq}) | \text{Cu(s)}$
 $E^\circ_{\text{Cu}^{2+}(\text{aq})/\text{Cu(s)}} = +0.34\text{V}$ কোষটির জন্য ইএমএফ এর মান কোনটি?
 (ক) 0.00V
 (খ) +0.34V
 (গ) -0.34V
 (ঘ) +0.17V

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S Marginson



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web annual leave in accordance with the employment act you are entitled to paid annual leave if you have worked for your employer for at least 3 months find out about your entitlements

request for one day time off form gov sg - Oct 23 2023

web request for one day time off 1 name of staff 2 form class 3 requested date for time off submit now

time off application form skss - Apr 17 2023

web to be submitted at least 3 working days in advance officers will receive approval non approval via email for urgent time off requests less than 3 working days please submit this form and also text meet principal personally 1 officer s name as in nric 2 officer s email address 3 time off date

hours of work overtime and rest day ministry of manpower - Mar 16 2023

web for overtime work your employer must pay you at least 1 5 times the hourly basic rate of pay payment must be made within 14 days after the last day of the salary period a non workman earns 2 600 a month and works 2 hours of overtime the overtime pay is 13 60 1 5 2 hours 40 80

time off application form gov sg - Jul 20 2023

web 1 please ensure approval is granted before taking your time off 2 inform the kps on duty for the arrangement of relief if required