Life Sciences

Grade 11

Term 3



Date: September 2024

Time: 2 hours

Total: 80

QUESTIONS	TYPE OF QUESTION	TOPIC	MARK ALLOCATION
G	Multiple choice	Animal mybition Human alimentary consi Wechanical digestion	19
u	Terminology	Appropriate and assemble or	
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SECTION A

Question 1

- 1.1 Different options are given as possible answers to the following questions. Choose the correct answer and write only the letter (A to D) next to the question number (1.1.1 to 1.1.10) on the FOLIO PAPER ex. 1.1.11 E.
- 1.1.1 Small digested nutrients are taken up into the bloodstream.
 - A. Ingestion
 - 5. Assimilation
 - C. Absorption
 - D. Egestion
- 1.1.2 The muscular tube that connects the mouth cavity to the stomach.
 - A. Desophagus
 - 8. Epiplottis
 - C. Traches
 - D. Larynx
- 1.1.3 The colon absorbs water from indigestible food in the alimentary canal.
 - A. Duodenum
 - 5. Ileum
 - C. Large intestine
 - D. Small intestine
- 1.1.4 The liver secretes...
 - A. bile that helps to break down fat
 - saliva that contains enzymes
 - C. stores bile from the liver before it is secreted into the duodenum of the small intestine-
 - D pancreatic juice that contains encymes

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared stoichiometry the reaction of iron with copper ii sulfate - Jun 13 2023

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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