

## Gas Stoichiometry Practice Sheet

- 1) For the reaction  $2 \text{H}_{2(g)} + \text{O}_{2(g)} \rightarrow 2 \text{H}_2\text{O}_{(g)}$ , how many liters of water can be made from 5 L of oxygen gas and an excess of hydrogen?
  
- 2) How many liters of water can be made from 55 grams of oxygen gas and an excess of hydrogen at STP?
  
- 3) How many liters of water can be made from 55 grams of oxygen gas and an excess of hydrogen at a pressure of 12.4 atm and a temperature of 85° C?
  
- 4) How many liters of water can be made from 34 grams of oxygen gas and 6.0 grams of hydrogen gas at STP? What is the limiting reactant for this reaction?

# Gas Stoichiometry Practice Sheet 2007 Cavalcade

**RJ Alexander**



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