



Use and Care Guide

Automatic Dryers

TEDL600W

Thank you for choosing an Estate appliance.

This Use and Care Guide will help you operate and maintain your new, quality-built automatic dryer. Keep this Use and Care Guide in a safe place for future reference.

Please complete and mail the Owner Registration Card provided with this product. Then complete the form below. Have this information ready if you need service or call with a question.

- Copy model and serial numbers from plate (see diagram on page 4) and purchase date from sales slip.
- Keep this book and sales slip together in a handy place.

Model Number

Serial Number

Puirchase Date

Service Company Phone Number

Please complete all of the Installation instruction

steps before attempting to operate your dryer.

You are responsible for

It is your responsibility to be sure that your driver:

- Is properly installed by a qualified person."
- Is installed and leveled on a floor that can support the weight of the dryer."
- Is located in a well ventilated room where the temperature is above 45°F (7.2°C).
- Is properly connected to a grounded electrical circuit that meets the electrical requirements.*
- Is properly connected to fuel and exhaust systems.
- Is not installed against drapes or curtains."
- Is used only for the job it was designed to do
- Is not used by children or anyone unable to operate it properly.
- Is properly maintained.
- See Installation Instructions for complete information.

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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid h 2 co 3 and its conjugate base the bicarbonate ion hco 3

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an acid and a base react not to produce a salt and a solvent but to form a new acid and a new base the concept of neutralization is thus absent brønsted lowry acid base behavior is formally independent of any solvent making it more all encompassing than the arrhenius model

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acids bases and enzymes many acids and bases in living things provide the ph that enzymes need enzymes are biological catalysts that must work effectively for biochemical reactions to occur most enzymes can do their job only at a certain level of acidity cells secrete acids and bases to maintain the proper ph for enzymes to do their work

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acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen

ion from an acid acidic substances are usually identified by their sour taste

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a binary acid is an acid that consists of hydrogen and one other element the most common binary acids contain a halogen the acid name begins with the prefix hydro followed by the base name of the anion followed by the suffix ic

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oct 18 2023 most of the acids are solid it has a sour taste they produce hydrogen after reacting with active metals like zinc magnesium aluminium or iron acids have a low ph value i e below 7 salts and water are produced when an acid reacts with base or alkalis most of the acids are solids

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arrhenius s definition of acids and bases the earliest definition of acids and bases is arrhenius s definition which states that an acid is a substance that forms hydrogen ions h when dissolved in water and a base is a substance that forms hydroxide ions oh when dissolved in water

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other s chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is ce acid base water salt nonumber

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