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3rd Edition

Chemical Bonds

2-1 If the isotope ^{32}S has 16 protons and 16 neutrons, how many protons and how many neutrons will the isotope ^{33}S have?

- 2-2** A. If 0.5 mole of glucose weighs 90 g, what is the molecular weight of glucose?
 B. What is the concentration, in grams per liter (g/l), of a 0.25 M solution of glucose?
 C. How many molecules are there in 1 mole of glucose?

2-3 Which of the following elements is LEAST abundant in living organisms?

- (a) Sulfur
- (b) Carbon
- (c) Oxygen
- (d) Nitrogen
- (e) Hydrogen

2-4 Your friend learns about Avogadro's number and thinks it is so huge that there may not even be a mole of living cells on Earth. You have recently heard that there are about 50 trillion (50×10^{12}) human cells in each adult human body, so you bet your friend \$5 that there is more than a mole of cells on Earth. Once you learn that each human contains more bacterial cells (in the digestive system) than human cells, you are sure that you have won the bet. The human population is now more than 6 billion (6×10^9). What calculation can you show your friend to convince him you are right?

2-5 Atoms form covalent bonds with each other by

- (a) sharing protons.
- (b) sharing electrons.
- (c) transferring electrons from one atom to the other.
- (d) sharing neutrons.
- (e) attraction of positive and negative charges.

2-6 A carbon atom contains six protons and six neutrons.

- A. What are its atomic number and atomic weight?
- B. How many electrons does it have?
- C. What is its valence? How does this affect carbon's chemical behavior?
- D. Carbon with an atomic weight of 14 is radioactive. How does it differ in structure from

Essential Cell Biology 3rd Edition Practice Questions

Rachel Sandford



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Essential Cell Biology Bruce Alberts, Dennis Bray, Karen Hopkin, Alexander D Johnson, Julian Lewis, Martin Raff, Keith Roberts, Peter Walter, 2015-01-01 Essential Cell Biology provides a readily accessible introduction to the central concepts of cell biology and its lively clear writing and exceptional illustrations make it the ideal textbook for a first course in both cell and molecular biology The text and figures are easy to follow accurate clear and engaging for the introductory student Molecular detail has been kept to a minimum in order to provide the reader with a cohesive conceptual framework for the basic science that underlies our current understanding of all of biology including the biomedical sciences The Fourth Edition has been thoroughly revised and covers the latest developments in this fast moving field yet retains the academic level and length of the previous edition The book is accompanied by a rich package of online student and instructor resources including over 130 narrated movies an expanded and updated Question Bank Essential Cell Biology Fourth Edition is additionally supported by the Garland Science Learning System This homework platform is designed to evaluate and improve student performance and allows instructors to select assignments on specific topics and review the performance of the entire class as well as individual students via the instructor dashboard Students receive immediate feedback on their mastery of the topics and will be better prepared for lectures and classroom discussions The user friendly system provides a convenient way to engage students while assessing progress Performance data can be used to tailor classroom discussion activities and lectures to address students needs precisely and efficiently For more information and sample material visit <http://garlandscience.rocketmix.com>

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Introduction to the Biology of Marine Life James L. Sumich, John Francis Morrissey, 2004 This textbook examines selected groups of marine organisms within a framework of basic biological principles and processes With attention to taxonomic evolutionary ecological behavioral and physiological aspects of biological study the book contains chapters on habitat patterns of association phytoplankton marine plants protozoans and inv

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