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Kathleen Armour



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web may 28 2020 answer problem 4.3.1 3.4.3.1.3 for each of the following compounds state whether it is ionic or covalent if it is ionic write the symbols for the ions involved a. NF_3 b. BaO c. NH_4Cl d. CO_2 e. H_2SO_4 f. Na_2O answer a. answer b. answer c. answer d. answer e. answer f. problem 4.3.1 4.4.3.1.4

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web iron is an element it is made of only one type of atom iron atoms sulfur is another element it contains only sulfur atoms and nothing else so it is pure when we mix two different pure

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web jul 25 2022 write the name for each molecular compound CO_2 H_2S SF_6 N_2O_2 strategy first you have to know how to name a molecular compound to name a compound its the prefix name of first element prefix and name of the second element

with the suffix ide the prefixes are as follows mono 1 di 2 tri 3 tetra 4

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web molecular compounds are inorganic compounds that take the form of discrete molecules examples include such familiar substances as water H_2O and carbon dioxide CO_2 these compounds are very different from ionic compounds like sodium chloride NaCl

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web may 28 2020 what is the total mass of hydrogen in each of the molecules a CH_4 b CHCl_3 c $\text{C}_{12}\text{H}_{10}\text{O}_6$ d $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ answer a answer b answer c answer d problem 4 4 1 3 calculate the molecular or formula mass of each of the following a P_4 b H_2O c $\text{Ca(NO}_3)_2$ d $\text{CH}_3\text{CO}_2\text{H}$ acetic acid