

(vi) Identify the compound 'B' in the following series of reaction :

$$\text{Propanenitrile} \xrightarrow[\text{dil. HCl}]{\text{Na / alc}} \text{A} \xrightarrow{\text{NaNO}_2} \text{B}$$

- (a) n-Propyl chloride
 (c) n-Propyl alcohol

- (b) Propanamine
 (d) Isopropyl alcohol

(vii) Which of the following reagents is best for the following conversion ?



- (a) LiAlH_4
 (b) H_2O
 (c) H_2/Ni , 453 K
 (d) $\text{Zn-Hg} + \text{HCl}(\text{conc.})$

Q. 6. Answer any SIX of the following :

[12]

- (i) Calculate magnetic moment of $\text{Fe}^{2+}_{(\text{aq})}$ ion ($Z = 26$)
- (ii) How is ethanol prepared from methanal by using Grignard reagent?
- (iii) Write the chemical reaction to prepare novolac polymer.
- (iv) Why does p-nitrochlorobenzene undergo displacement reactions readily with attack of nucleophilic HO^- ion?
- (v) What is the action of bromine in alkaline medium on –
 (a) $\text{CH}_3\text{CH}_2\text{NO}_2$ (b) $\text{CH}_3 - \underset{\text{CH}_3}{\text{CH}} - \text{NO}_2$

(vi) Define antioxidants and mention two examples.

(vii) How is 4-methylpent-3-en-2-one obtained from propan-2-one?

(viii) What are hormones? Write the structure of simple triglycerides

Q. 7. Answer any THREE of the following:

[9]

- (i) Write the different oxidation states of manganese. Why + 2 oxidation state of manganese is more stable ?
- (ii) How are the following compounds prepared ?
 (a) Benzaldehyde from benzene (b) Acetophenone from benzene
 (c) Benzaldehyde from benzoyl chloride
- (iii) Define complex lipids and write the structures of nucleotide and nucleoside.
- (iv) Write the formulae of the following compounds –
 (a) Sodium hexanitrito – N - cobaltate (III)
 (b) Tetraaquodichlorochromium (III) chloride
 (c) Potassium tetracyanoaurate (III) ion

Q. 8. Answer any ONE of the following:

[7]

- (1) (a) Explain the following terms : (i) Homopolymers (ii) Elastomers
 (b) Explain the mechanism of cleansing action of soaps.
 (c) Write balanced chemical equations for the action of –
 (i) phosphorous trichloride on propan – 2 – ol
 (ii) hydrogen bromide on styrene in the presence of a peroxide
 (iii) methyl bromide on silver propanoate
- (2) (a) Write a short note on Hoffmann bromamide degradation
 (b) Explain the mechanism of action of hydroiodic acid on 3-methylbutan-2-ol
 (c) Mention 'two' uses of propan-2-one

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