

CHEMISTRY: LIMITING REAGENT Practice Problems

Please show your work on a separate piece of paper!

Assume standard temperature and pressure, STP conditions for all reactions.

1. At high temperatures, sulfur combines with iron to form the brown-black iron (II) sulfide:



In one experiment, 7.62 g of Fe are allowed to react with 8.67 g of S.

- What is the limiting reagent, and what is the reactant in excess?
- Calculate the mass of FeS formed.

2. Acrylonitrile, $\text{C}_3\text{H}_3\text{N}$, is the starting material for the production of a kind of synthetic fiber (acrylics) and can be made from propylene, C_3H_6 , by reaction with nitric oxide, NO , as follows:



What mass of $\text{C}_3\text{H}_3\text{N}$ can be made when 21.6 g of C_3H_6 react with 21.6 g of nitric oxide?

3. Calculate the percent yield for the reaction: $\text{P}_{4(s)} + 6 \text{Cl}_{2(g)} \rightarrow 4 \text{PCl}_{3(l)}$
if 75.0 g of phosphorus reacts with excess chlorine gas to produce 111.0 g of phosphorus trichloride.

4. Formic acid, HCHO_2 , burns in oxygen to form carbon dioxide and water as follows:



If a 3.15-g sample of formic acid was burned in 2.0 L of oxygen, what volume of carbon dioxide would be produced?

5. Zinc metal reacts with hydrochloric acid to produce zinc chloride and hydrogen gas.
- Balance the following reaction: $\text{Zn}_{(s)} + \text{HCl}_{(aq)} \rightarrow \text{ZnCl}_{2(aq)} + \text{H}_{2(g)}$
 - A 3.50-g sample of zinc metal is allowed to react with 2.50 g of hydrochloric acid.

Complete the following table:

reactants/products	Zn (grams)	HCl (grams)	ZnCl ₂ (grams)	H ₂ (L)
Before reaction				
After reaction	1.26 g			

Limiting Reactant Problems Answer Key

LP Steffe



Limiting Reactant Problems Answer Key:

This book delves into Limiting Reactant Problems Answer Key. Limiting Reactant Problems Answer Key is a vital topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Limiting Reactant Problems Answer Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Limiting Reactant Problems Answer Key
 - Chapter 2: Essential Elements of Limiting Reactant Problems Answer Key
 - Chapter 3: Limiting Reactant Problems Answer Key in Everyday Life
 - Chapter 4: Limiting Reactant Problems Answer Key in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, this book will provide an overview of Limiting Reactant Problems Answer Key. This chapter will explore what Limiting Reactant Problems Answer Key is, why Limiting Reactant Problems Answer Key is vital, and how to effectively learn about Limiting Reactant Problems Answer Key.
 3. In chapter 2, the author will delve into the foundational concepts of Limiting Reactant Problems Answer Key. This chapter will elucidate the essential principles that need to be understood to grasp Limiting Reactant Problems Answer Key in its entirety.
 4. In chapter 3, the author will examine the practical applications of Limiting Reactant Problems Answer Key in daily life. This chapter will showcase real-world examples of how Limiting Reactant Problems Answer Key can be effectively utilized in everyday scenarios.
 5. In chapter 4, the author will scrutinize the relevance of Limiting Reactant Problems Answer Key in specific contexts. The fourth chapter will explore how Limiting Reactant Problems Answer Key is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Limiting Reactant Problems Answer Key. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Limiting Reactant Problems Answer Key.

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