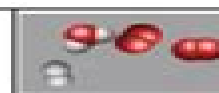


**Activity A:****Water reaction**Get the Gizmo ready:

- Make sure  $\text{H}_2 + \text{O}_2$  becomes  $\text{H}_2\text{O}$  is selected.
- Click **Reset** (↺).

**Goal: Identify a limiting reactant**

1. Count:  $\text{H}_2\text{O}$  is the **chemical formula** for water. In order to produce a single molecule of water, how many hydrogen atoms are needed? 2 hydrogen atoms Oxygen atoms? 1 oxygen atom

2. Predict: Set the number of  $\text{O}_2$  molecules to five and the number of  $\text{H}_2$  molecules to eight.

- A. How many oxygen atoms are present? 10 atoms Hydrogen atoms? 16 atoms
- B. How many water molecules could form from these reactants? 8 molecules
- C. After the molecules react, which reactant will be left over? 8
- D. Which reactant will be the limiting reactant? Oxygen
- E. Click **Play** and wait until **Reaction complete** is shown. What happened?

Due to the hydrogen and oxygen reacting, there are eight water molecules and one remaining oxygen molecule.

3. Test: Click **Reset**. Set the number of  $\text{O}_2$  molecules to 10.

- A. How many hydrogen molecules ( $\text{H}_2$ ) will be needed for there to be no excess reactants?  Use the slider to set the number of  $\text{H}_2$  molecules to this value.  
20 hydrogen molecules

- B. How many water molecules will be formed? 20
- C. Click **Play**. What happened? 20 water molecules were formulated

4. Challenge yourself: Run the Gizmo with five  $\text{O}_2$  molecules and nine  $\text{H}_2$  molecules.

- A. How many of each molecule do you think will be produced by this reaction?

$\text{O}_2$ : 4 used 1 left  $\text{H}_2$ : 8 used 1 left  $\text{H}_2\text{O}$ : 8

- B. Click **Play**. How many were actually produced?  $\text{O}_2$ :

4  
use  
d 1

$\text{H}_2$ :

8  
use  
d 1

$\text{H}_2\text{O}$ :

8

# Limiting Reactant Gizmo Answer

**Lifelique**



**Limiting Reactant Gizmo Answer:**

Limiting Reactant and Percent Lifeliqe,2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction

## Reviewing **Limiting Reactant Gizmo Answer**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Limiting Reactant Gizmo Answer**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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