Complete the following questions. Include units and chemical symbols in your answer. Use amu to the tenths.

$$Zn_{(s)} + 2 HCI_{(aq)} \rightarrow ZnCI_{2(s)} + H_{2(g)}$$

- Using the above reaction, what is the limiting reactant when 5.6 g zinc is added to a solution that contains 23.0 g of hydrochloric acid? (6 pts)
- In the above reaction, is there excess reagent? If so, how much? (2 pts)
- If 26.3 g of silicon dioxide is heated with an excess of carbon, 16.2 g
 silicon carbide is produced. What is the percent yield of this reaction? (7 pts)

$$SiO_2(s) + 3C(s) \rightarrow SiC(s) + 2CO(g)$$

Classify each of the following as either: synthesis(combination), decomposition, single replacement, double replacement, or combustion.

Limiting Reactant And Percentage Yield Quiz

Jeffrey Gaffney, Nancy Marley

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