

Concept Review

Section: Limiting Reactants and Percentage Yield

Complete each statement below by choosing a term from the following list. Some may be used more than once.

reactant product limiting actual yield
 percentage yield excess theoretical

1. The _____ is the substance that is completely used up in a chemical reaction.
2. The _____ is the substance that is not completely used in a chemical reaction.
3. The amount that is expected to be produced is the _____ yield.
4. The actual amount of product that is produced is the _____ yield.
5. The ratio of the actual yield to the theoretical yield is the _____ yield.
6. The amount of product that is actually produced is the _____ yield.
7. The _____ is the amount of product that is actually produced.
8. The _____ is the amount of product that is expected to be produced.
9. The _____ is the amount of product that is actually produced.
10. The _____ is the amount of product that is actually produced.
11. The _____ is the amount of product that is actually produced.
12. The _____ is the amount of product that is actually produced.
13. The _____ is the amount of product that is actually produced.
14. The _____ is the amount of product that is actually produced.
15. The _____ is the amount of product that is actually produced.

Answer the following items in the space provided.

16. Write the balanced equation for the reaction of hydrogen gas with oxygen gas to form water. What is the theoretical yield of water if 2.00 g of hydrogen gas reacts with 16.00 g of oxygen gas?



Limiting Reactants And Percent Yield 9 3 Section Review

Jianjun Gao



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