



Stoichiometry Question

Consider the synthesis of water produced in the Model. The reaction container is filled with 10.0 g of H_2 and 1.0 g of O_2 .

- a. Which reactant is the degree of progress in the limiting reactant in this case? Show your work. Also indicate that you are given standard quantities in each case (they are not given).

$$10.0 \text{ g } H_2 \left(\frac{1 \text{ mol } H_2}{2.016 \text{ g}} \right) = 4.96 \text{ mol } H_2$$

$$1.0 \text{ g } O_2 \left(\frac{1 \text{ mol } O_2}{32.00 \text{ g}} \right) = 0.031 \text{ mol } O_2$$

	$2 H_2 \rightarrow$	$1 O_2$	$\rightarrow 2 H_2O$
Before	4.96 mol 	0.031 mol	0 mol
Change 10% to 0.31g for 1.0g for 1.0g for 1.0g	-0.31 -0.31 -0.31 -0.31	-0.31 -0.31 -0.31 -0.31	+0.62 +0.62 +0.62 +0.62
After	4.65 mol 	0 mol	0.62 mol

Change in the limiting reactant



- b. What mass of water can be produced? Show your work.

$$0.031 \text{ mol } O_2 \left(\frac{2 \text{ mol } H_2O}{1 \text{ mol } O_2} \right) = 0.062 \text{ mol } H_2O$$

- c. Which reactant is present in excess, and what mass of that reactant remains after the reaction is complete?

H_2 is present in excess
initial mass of H_2 remains

$$4.96 \text{ mol } H_2 \left(\frac{2 \text{ mol } H_2O}{1 \text{ mol } O_2} \right) = 9.92 \text{ mol } H_2O$$

Limited And Excess Reactants Pogil Answers

SA Adler



Limited And Excess Reactants Pogil Answers:

Limited And Excess Reactants Pogil Answers Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Limited And Excess Reactants Pogil Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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