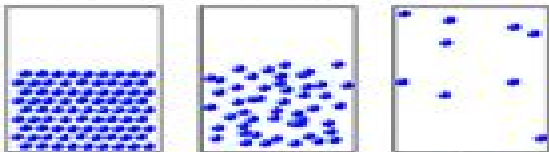


Kinetic Molecular Theory Worksheet

1. The word *kinetic* comes from a Greek word that means "to move." The **kinetic molecular theory** is based upon the assumption that particles of matter (atoms or molecules) are in constant **random motion**.
 2. Of the three states of matter, which one has the most kinetic energy? **Gas**.
 3. Which state of matter has particles that are separated by the largest distance? **Gas**.
- 

The diagram consists of three rectangular boxes labeled 'Solid', 'Liquid', and 'Gas' below them. The 'Solid' box is filled with a dense, regular grid of blue dots. The 'Liquid' box is filled with blue dots in a disordered, close-packed arrangement. The 'Gas' box contains only a few blue dots widely spaced throughout the container.
4. A scientific theory is an explanation of some type of natural phenomena. Theories are normally developed from careful study of the way the world behaves. Let's look at how gases behave and see if the kinetic molecular theory makes sense.
 5. Compared to liquids and solids, gases tend to have **lower** densities. This can be explained because the particles of gas are **in constant motion**.
 6. If you apply pressure to a sample of gas, it is fairly easy to **compress** its volume (think about what would happen to a balloon if you squeeze it gently). This can be explained because there is a lot of **empty space** between gas particles.
 7. If someone sprays perfume in one corner of the room, eventually a person on the other side of the room can smell it. This can be explained because gas particles move **quick** and **random**. In general, we would expect lighter gas particles to travel **faster** than heavier gas particles.
 8. When two gases mix together or move through each other, this process is known as **diffusion**. When gas particles escape out of a tiny hole in a container, this process is known as **effusion**. You should know the difference between these two words so you can avoid any confusion!
 9. Kinetic molecular theory can be summarized as follows:
 - a. Gas particles are in constant **motion**.
 - b. Gas particles are separated by relatively **large** distances.
 - c. When gas particles collide, they **do not transfer** kinetic energy.
 - d. Gas particles have **no** attractive or repulsive forces between them.
 - e. The kinetic energy of a gas is dependent on the **temperature** of the gas.

Kinetic Molecular Theory Answer Key Chemistry

SA Dillow



Kinetic Molecular Theory Answer Key Chemistry:

Decoding **Kinetic Molecular Theory Answer Key Chemistry**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Kinetic Molecular Theory Answer Key Chemistry**," a mesmerizing literary creation penned by a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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