

Edexcel iGCSE Chemistry 4CH0 Learning Plan

Section A2: Atomic structure		
Specification	Resources	Assessment
1.8 recall that atoms consist of a central nucleus, composed of protons and neutrons, surrounded by electrons, orbiting in shells 1.9 recall the relative mass and relative charge of a proton, neutron and electron 1.10 understand the terms atomic number, mass number, isotopes and relative atomic mass (A_r) 1.11 calculate the relative atomic mass of an element from the relative abundances of its isotopes 1.12 understand that the Periodic Table is an arrangement of elements in order of atomic number 1.13 deduce the electronic configurations of the first 20 elements from their positions in the Periodic Table 1.14 deduce the number of outer electrons in a main group element from its position in the Periodic Table.	Video: Section 1 – Lesson 1 – States of Matter and Atomic Structure (from 18:55 to the end). Powerpoint: Chemistry - Section 1 – Lesson 1 (slide 55 to end) Textbook: Page 6 – Chapter 2: Atomic Structure Page 7 – The structure of the atom Page 7 – Atomic number and mass number Page 8 – The electrons; to Page 11 Page 12 – Chapter check list	Textbook Page 12 – end of chapter questions. Talking paper – Edexcel Chemistry Section A2 Atomic structure. Section A2 Question – Atomic structure. (pdf) Section A2 Question – mark scheme. (pdf)

Videos – www.igcse-sciencecourses.com

Textbook Ref: Edexcel International GCSE Chemistry Student Book - Clark

DVD Video Clips – see resource DVD in textbook.

Gcse Chemistry 2012 Jac 1c

S Ben Porath

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Table of Contents Gcse Chemistry 2012 Jac 1c

1. Understanding the eBook Gcse Chemistry 2012 Jac 1c
 - The Rise of Digital Reading Gcse Chemistry 2012 Jac 1c
 - Advantages of eBooks Over Traditional Books
2. Identifying Gcse Chemistry 2012 Jac 1c
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gcse Chemistry 2012 Jac 1c
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gcse Chemistry 2012 Jac 1c
 - Personalized Recommendations
 - Gcse Chemistry 2012 Jac 1c User Reviews and Ratings

- Gcse Chemistry 2012 Jac 1c and Bestseller Lists
- 5. Accessing Gcse Chemistry 2012 Jac 1c Free and Paid eBooks
 - Gcse Chemistry 2012 Jac 1c Public Domain eBooks
 - Gcse Chemistry 2012 Jac 1c eBook Subscription Services
 - Gcse Chemistry 2012 Jac 1c Budget-Friendly Options
- 6. Navigating Gcse Chemistry 2012 Jac 1c eBook Formats
 - ePub, PDF, MOBI, and More
 - Gcse Chemistry 2012 Jac 1c Compatibility with Devices
 - Gcse Chemistry 2012 Jac 1c Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gcse Chemistry 2012 Jac 1c
 - Highlighting and Note-Taking Gcse Chemistry 2012 Jac 1c
 - Interactive Elements Gcse Chemistry 2012 Jac 1c
- 8. Staying Engaged with Gcse Chemistry 2012 Jac 1c
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gcse Chemistry 2012 Jac 1c
- 9. Balancing eBooks and Physical Books Gcse Chemistry 2012 Jac 1c
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gcse Chemistry 2012 Jac 1c
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gcse Chemistry 2012 Jac 1c
 - Setting Reading Goals Gcse Chemistry 2012 Jac 1c
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gcse Chemistry 2012 Jac 1c
 - Fact-Checking eBook Content of Gcse Chemistry 2012 Jac 1c
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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