



# Hitachi Rc1205 Manual

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web compiled by g izquierdo rodríguez page 2 1 directly proportional to the product of their charges  $f \propto q_1 q_2$  2 inversely proportional to square of the distance between them  $f \propto \frac{1}{r^2}$  therefore  $f \propto \frac{q_1 q_2}{r^2}$  coulomb s law in symbols  $f = k \frac{q_1 q_2}{r^2}$  coulomb s law the magnitude of the electrostatic force exerted by one point charge  $q$

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web choose 1 answer the solution becomes negatively charged due to the majority  $\text{Cl}^-$  ions a the solution becomes negatively charged due to the majority  $\text{Cl}^-$  ions the solution becomes positively charged due to the stronger  $\text{Mg}^{2+}$  ions b the solution becomes positively charged due to the stronger  $\text{Mg}^{2+}$  ions the solution remains neutrally

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web answer d somewhere in region d justification with the test charge and c 1 being negative there is a repulsive force on the test charge to the right from c 2 there is an attractive force on the test charge to the left by referring to coulomb's law  $F = k \frac{q_1 q_2}{r^2}$  1  $q_1 q_2 < 0$  2 we know that the force

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 dextrose and or lipids tpn refers to amino acids dextrose and fat emulsion if a drug is stated compatible in tpn it

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 dextrose electrolytes minerals trace elements vitamins and sometimes other additives drugs such as heparin

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 trophamine initiation 1 5 2 gm kg day advancement 0 5 1 gm kg day goal 1 5 3 gm kg day initiation 1 2 gm kg day  
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