

F321: Atoms, Bonds and Groups

Structure & Bonding – Mark Scheme

1. (i) (Electrostatic) **attraction** between oppositely charged **ions** ✓

IGNORE force

IGNORE references to transfer of electrons

MUST be ions, not particles

1

- (ii) Mg shown with either 8 or 0 electrons

AND

S shown with 8 electrons **with** 2 crosses and 6 dots (or vice versa) ✓

Correct charges on both ions ✓



Mark charges on ions and electrons independently

For first mark, if 8 electrons are shown around the Mg then 'extra electrons' around S must match the symbol chosen for electrons around Mg

Shell circles not required

IGNORE inner shell electrons

Brackets are not required

2

[2]

2. (i) Electron pairs in covalent bonds shown correctly using dots and crosses in a molecule of the F_2O ✓

Lone pairs correct on O and both F atoms ✓



Must be 'dot-and-cross'

circles for outer shells NOT needed

IGNORE inner shells

Non-bonding electrons of O do not need to be shown as pairs

Non-bonding electrons of F do not need to be shown as pairs

2

F321 Chemistry June 2013 Ms

CL Gary



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Decoding **F321 Chemistry June 2013 Ms**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**F321 Chemistry June 2013 Ms**," a mesmerizing literary creation penned with a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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