F321: Atoms, Bonds and Groups Structure & Bonding – Mark Scheme

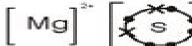
1. (i) (Electrostatic) attraction between oppositely charged ions -

IGNORE force
IGNORE references to transfer of electrons
MUST be tons, not particles

(ii) Mg shown with either 8 of 0 electrons

S shown with 8 electrons with 2 crosses and 6 dots (or vice versa) 🗸

Correct charges on both ions



Mark charges on tons and electrons independently

For first mark, if 6 electrons are shown around the Mg then

'extra electrons' around S must match the symbol chosen for

electrons around Mg Stell circles not reasured

IGNORE taner shell electrons

Brackets are not required

 (i) Electron pairs in covalent bonds shown correctly using dots and crosses in a molecule of the F₂O

Lone pairs correct on O and both F atoms 🛩



Must be 'dot-and-cross' arcles for outer shells NOT needed IGNORE sneer shells

Non-bonding electrons of O do not need to be shown as pairs Non-bonding electrons of F do not need to be shown as pairs [27]

1

2

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CL Gary



Decoding F321 Chemistry June 2013 Ms: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "F321 Chemistry June 2013 Ms," a mesmerizing literary creation penned with a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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