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$$3 \text{ KClO}_3 \rightarrow 3 \text{ KCl} + 3 \text{ O}_2$$

$$2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF}$$

$$1 \text{ Cl}_2 + 2 \text{ H}_2 \rightarrow 2 \text{ HCl}$$

$$2 \text{ H}_2\text{O} + 5 \text{ Pb} \rightarrow 2 \text{ H}_2 + 5 \text{ PbO}$$

$$2 \text{ HCl} + 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2 \rightarrow 2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 + 6 \text{ KBr} + 1 \text{ Al}_2\text{SO}_4 + 3 \text{ H}_2 + 1 \text{ CH}_4$$
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$$1 \text{ N}_2 + 2 \text{ F}_2 \rightarrow 2 \text{ NF}_3$$

$$2 \text{ C}_6\text{H}_{10} + \text{O}_2 \rightarrow 2 \text{ CO}_2 + 2 \text{ H}_2\text{O}$$

$$3 \text{ HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$$

$$\text{GaBr} + \text{NaBr} \rightarrow \text{SnO}_4 + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{S}_3 + 5 \text{ NF}_3 + \text{SnF}_2 + \text{N}_2\text{O}_3$$

using the equation from problem 2 above answer the following

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$$1 \text{ N}_2 + 3 \text{ F}_2 \rightarrow 2 \text{ NF}_3$$

$$2 \text{ C}_6\text{H}_{10} + 17 \text{ O}_2 \rightarrow 12 \text{ CO}_2 + 10 \text{ H}_2\text{O}$$

co2 10 h2o 3 4 5 1 hbr 1 khco3 1 h2o 1 kbr 1 co2 2 gabr3 3 na2so3 1 ga2 so3 3 6 nabr 3 sno 2 nf3 3 snf2 1 n2o3 using the following equation 2 naoh h2so4 2 h2o na2so4

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