

## Mole and Stoichiometry Mini-Lab

7. Write and balance the equation for the reaction between aluminum and water to produce aluminum oxide and hydrogen gas.



8. Using the moles of the aluminum can in #1 as the GIVEN, how many moles of hydrogen gas can be produced? Assume water is available in excess. UNKNOWN: mol  $\text{H}_2$  GIVEN: mol Al (answer to #1)

9. Using the moles of water from #4 as the GIVEN with aluminum in excess, calculate the number of moles of aluminum oxide that can be produced.



## Mini Lab 1 apply Stoichiometry

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## **Mini Lab 1 apply Stoichiometry:**

## Mini Lab 1apply Stoichiometry Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has become more evident than ever. They have the capability to inspire, provoke, and ignite change. Such could be the essence of the book **Mini Lab 1apply Stoichiometry**, a literary masterpiece that delves deep in to the significance of words and their affect our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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