

21. In which of the following reactions addition does not take place according to Markovnikov's rule ? (IIT 2000)
- (a)  $\text{CH}_3\text{CH}=\text{CH}_2 + \text{HCl} \xrightarrow{\text{ROOR}}$   
 (b)  $\text{CH}_3\text{CH}=\text{CH}_2 + \text{HBr} \xrightarrow{\text{ROOR}}$   
 (c)  $\text{CH}_2=\text{CH}_2 + \text{HI} \xrightarrow{\text{ROOR}}$   
 (d)  $\text{CH}_3\text{CH}=\text{CHCH}_3 + \text{HBr} \xrightarrow{\text{ROOR}}$
22. Ozonolysis of 2-methyl butene-2 yields : (UPC 2003)
- (a) ketone (b) aldehyde  
 (c) aldehyde & ketone (d) none
23.  $\text{C}_6\text{H}_{12}$  on ozonolysis gave only propanone. Alkene is : (AFMC 2003)
- (a)  $\text{CH}_3\text{CH}_2\text{CH}=\text{C}(\text{CH}_3)_2$   
 (b)  $(\text{CH}_3)_2\text{C}=\text{CH}-\text{CH}_3$   
 (c)  $(\text{CH}_3)_2\text{C}=\text{C}(\text{CH}_3)_2$   
 (d)  $\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_2\text{CH}_3$
24. The reaction,  
 $\text{CH}_2=\text{CHCH}_3 + \text{HBr} \longrightarrow \text{CH}_3\text{CHBrCH}_3$  is : (AIIMS 2003)
- (a) Nucleophilic addition  
 (b) Electrophilic addition  
 (c) Electrophilic substitution  
 (d) Free radical addition
25. Which of the following alkenes is most stable ? (UPC 2003)
- (a)  $\text{R}_2\text{C}=\text{CR}_2$  (b)  $\text{R}-\text{CH}=\text{CH}-\text{R}$   
 (c)  $\text{RCH}=\text{CH}_2$  (d)  $\text{CH}_2=\text{CH}_2$
26. A molecule of HCl adds on 3-methyl but-1-ene. Product formed is : (IIT 2001)
- (a)  $\text{CH}_3\text{CH}(\text{CH}_3)-\text{CHCl}-\text{CH}_3$   
 (b)  $\text{CH}_3-\text{C}(\text{Cl})(\text{CH}_3)-\text{CH}_2-\text{CH}_3$   
 (c)  $\text{CH}_3\text{CH}(\text{CH}_3)-\text{CH}_2-\text{CH}_2\text{Cl}$   
 (d)  $\text{CH}_3\text{CH}(\text{CH}_3)-\text{CH}(\text{Cl})-\text{CH}_2\text{CH}_3$
27. The intermediate during the addition of HCl to propene in presence of peroxide is : (IIT 2002)
- (a)  $\text{CH}_3\dot{\text{C}}\text{HCH}_2\text{Cl}$  (b)  $\text{CH}_3\dot{\text{C}}\text{HCH}_3$   
 (c)  $\text{CH}_3\text{CH}_2\dot{\text{C}}\text{H}_2$  (d)  $\text{CH}_3\text{CH}_2\dot{\text{C}}\text{H}$
28. Which of the following exhibit *cis, trans*-isomerism ?
- (a)  $\text{CH}_2\text{Br}-\text{CH}_2\text{Br}$  (b)  $\text{CBr}_2=\text{CH}_2$   
 (c)  $\text{CHBr}=\text{CHBr}$  (d)  $\text{CBr}_2=\text{CH}-\text{CH}_3$
29. The compound that dehydrates most readily is : (CPMT 2001)
- (a)  $\text{R}_2\text{CHOH}$  (b)  $\text{RCH}_2\text{OH}$   
 (c)  $\text{R}_3\text{COH}$  (d)  $\text{R}-\text{OH}$
30. The addition of propene with HOCl proceeds via the addition of : (IIT 2001)
- (a)  $\text{H}^+$  in the first step  
 (b)  $\text{Cl}^+$  in the first step  
 (c)  $\text{OH}^-$  in the first step  
 (d)  $\text{Cl}^+$  and  $\text{OH}^-$  in a single step
31. Which compound is sure to be alkene ? (AFMC 2003)
- (a)  $\text{C}_2\text{H}_4$  (b)  $\text{C}_3\text{H}_6$   
 (c)  $\text{C}_4\text{H}_8$  (d)  $\text{C}_5\text{H}_{10}$   
 (e) All
32. 1-chlorobutane with alc. KOH gives : (AFMC 2001)
- (a) 1-butene (b) 1-butanol  
 (c) 2-butene (d) 2-butanol
33. How much ethylene is oxidised by 3 litres of  $\text{O}_2$  under identical conditions ? [CBSE (Med.) 2002]
- (a) One mole (b) One litre  
 (c) 3 litre (d) 0.3 litre
34. Which one of the following alkenes will react fastest with  $\text{H}_2$  under catalytic hydrogenation condition ? [IIT (S) 2000]
- (a)  $\text{R}_2\text{C}=\text{CH}_2$  (b)  $\text{RHC}=\text{CHR}$   
 (c)  $\text{R}_2\text{C}=\text{CHR}$  (d)  $\text{R}_2\text{C}=\text{CR}_2$
35. Which of the following gases gives a red precipitate with ammonical solution of cuprous chloride ? (DCE 2004)
- (a)  $\text{C}_2\text{H}_4$  (b)  $\text{C}_2\text{H}_6$   
 (c)  $\text{C}_2\text{H}_2$  (d)  $\text{CH}_3\text{CH}_2\text{CH}_3$
36. Which will not give a precipitate with Tollen's reagent? (DCE 2002)
- (a) Ethyne  
 (b) 2-pentyne  
 (c) 1-butyne  
 (d) 3-methyl-1-butyne
37. Which hydrocarbon reacts with sodium in liquid  $\text{NH}_3$ ? (AIEEE 2008)
- (a)  $\text{CH}_3-\text{CH}_2-\text{C}\equiv\text{CH}$   
 (b)  $\text{CH}_3-\text{C}\equiv\text{C}-\text{CH}_3$   
 (c)  $\text{CH}_3-\text{CH}=\text{CH}-\text{CH}_3$   
 (d)  $\text{CH}_3-\text{CH}_2-\text{CH}=\text{CH}_2$
38. Increasing order of volatility of  $\text{C}_2\text{H}_6$ ,  $\text{C}_2\text{H}_4$ ,  $\text{C}_2\text{H}_2$  and  $\text{C}_6\text{H}_6$  is : (BHU 2003)

# Mcqs Of Hydrocarbons

**K Morrison**



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