

Unit 10



UNIT 10: PERIODIC TABLE: CONCEPTS AND APPLICATIONS

Section 10.1: Introduction

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

- _____ 10. A column in the periodic table
- _____ 10. A row in the periodic table
- _____ 20. Group 1 elements
- _____ 20. Elements that are shiny and conduct electricity
- _____ 20. Group 8 elements

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, or if the statement is false, change the underlined word or phrase to make it true.

- _____ 20. There are two representative elements of elements.
- _____ 20. Most elements are metals in the periodic table.
- _____ 20. Group 1 elements except for hydrogen are metals in the periodic table.
- _____ 20. Group 2 elements are the alkaline earth metals.
- _____ 20. Group 1 elements are highly reactive elements known as alkali metals.
- _____ 20. Group 8 elements are very reactive elements known as noble gases.
- _____ 20. Alkaloids have properties of both metals and non-metals.

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SB Merriam

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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