

SECTION - A

Q. 1. Select and write the correct answer for the following multiple choice type of questions :

[10]

(i) The product obtained in the following reaction

P 239
11.2



- (a) Pent-3-en-1-ol (b) ☒ Pentan-1-ol
(c) Pentan-2-ol (d) Pentanal

(ii) Amongst the following, the solubility of which ionic solid decreases with increase in temperature?

P 30
11)

- (a) KNO_3 (b) NaBr
(c) ☒ Na_2SO_4 (d) KCl

(iii) The correct IUPAC name of $\text{Na}_3 [\text{AlF}_6]$ is

P 196
Table
9.3

- (a) ☒ Sodium hexafluoroaluminate (III)
(b) Sodium hexafluoroaluminate (II)
(c) Sodium hexafluoroaluminium (III)
(d) Sodium hexafluoroaluminium (II)

(iv) Which of the following acids has highest pK_a value?

P 277
Table
12.9

- (a) Mono chloroacetic acid (b) Dichloroacetic acid
(c) Trichloroacetic acid (d) ☒ Acetic acid

(v) Number of carbon atoms present in isoprene unit is ____.

P 322
15.3

- (a) 6 (b) ☒ 5
(c) 4 (d) 3

(vi) The colourless transition metal ion amongst the following is

P 173
3d¹⁰

- (a) ☒ Cu^+ (b) Cu^{++}
(c) Ni^{++} (d) Co^{++}

Hsc Exam Chemistry Answer Paper

Jim Stamell

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Hsc Exam Chemistry Answer Paper:

Excel HSC Chemistry Jim Stamell, 2011 ISBN 9781741252996 AUTHOR Jim Stamell RRP 39 95 PAGES 428 pp
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