

Note:

- All questions are compulsory.
- Answer to the two sections are to be written in the same answer book.
- Figure to the right hand side indicate full marks.
- Write balanced chemical equations and draw neat and labelled diagrams wherever necessary.
- Every new question must be started on a new page.
- Use of logarithmic table is allowed.

CHEMISTRY: SECTION – I

Q. 1. Select and write the most appropriate answer from the given alternatives for each sub-question: [7]

- In body centred cubic structure the space occupied is about
(A) 68 % (B) 53 %
(C) 38 % (D) 32 %
- For a gaseous reaction the unit of rate of reaction is
(A) L atm s^{-1} (B) $\text{atm mol}^{-1} \text{s}^{-1}$
(C) atm s^{-1} (D) mol s
- Which of the following compounds contains S – O as well as S – S bonds?
(A) Sulphuric acid (B) Thiosulphuric acid
(C) Sulphurous acid (D) Thiosulphurous acid
- Which of the following solutions shows maximum depression in freezing point?
(A) 0.5 M Li_2SO_4 (B) 1 M NaCl
(C) 0.5 M $\text{Al}_2(\text{SO}_4)_3$ (D) 0.5 M BaCl_2
- For a chemical reaction $\Delta S = -0.035 \text{ kJ/K}$ and $\Delta H = -20 \text{ kJ}$.
(A) 5.14 K (B) 57.14 K
(C) 571.4 K (D) 5714.0 K
- The standard e.m.f of the following cell is 0.463 V
 $\text{Cu} | \text{Cu}_{(0.01)}^{++} || \text{Ag}_{(0.01)}^{+} | \text{Ag}$. If $E_{\text{Ag}}^{\circ} = 0.800 \text{ V}$,
What is the standard potential of Cu electrode?
(A) 1.137 V (B) 0.337 V
(C) 0.463 V (D) -0.463 V
- Fe_2O_3 is reduced to spongy iron near the top of blast furnace by
(A) H_2 (B) CaO
(C) SiO_2 (D) CO

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Karin Nielsen-Saines



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 - The Rise of Digital Reading Hsc Chemistry 2013board Question Paper Solution
 - Advantages of eBooks Over Traditional Books
2. Identifying Hsc Chemistry 2013board Question Paper Solution
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hsc Chemistry 2013board Question Paper Solution
 - User-Friendly Interface
4. Exploring eBook Recommendations from Hsc Chemistry 2013board Question Paper Solution
 - Personalized Recommendations
 - Hsc Chemistry 2013board Question Paper Solution User Reviews and Ratings
 - Hsc Chemistry 2013board Question Paper Solution and Bestseller Lists
5. Accessing Hsc Chemistry 2013board Question Paper Solution Free and Paid eBooks

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 - Hsc Chemistry 2013board Question Paper Solution Budget-Friendly Options
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- ePub, PDF, MOBI, and More
 - Hsc Chemistry 2013board Question Paper Solution Compatibility with Devices
 - Hsc Chemistry 2013board Question Paper Solution Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Hsc Chemistry 2013board Question Paper Solution
 - Highlighting and Note-Taking Hsc Chemistry 2013board Question Paper Solution
 - Interactive Elements Hsc Chemistry 2013board Question Paper Solution
8. Staying Engaged with Hsc Chemistry 2013board Question Paper Solution
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hsc Chemistry 2013board Question Paper Solution
9. Balancing eBooks and Physical Books Hsc Chemistry 2013board Question Paper Solution
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hsc Chemistry 2013board Question Paper Solution
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Hsc Chemistry 2013board Question Paper Solution
- Setting Reading Goals Hsc Chemistry 2013board Question Paper Solution
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Hsc Chemistry 2013board Question Paper Solution
- Fact-Checking eBook Content of Hsc Chemistry 2013board Question Paper Solution
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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