

## Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 1

**Q1.** At the same temperature and pressure, an equal volume of gases contains the same number of \_\_\_\_.

- a.) Molecules
- b.) Electrons
- c.) Protons
- d.) Particles

**Q2.** The gas law that relates the volume of a gas to the number of molecules of the gas is \_\_\_\_.

- a.) Gay-Lussac's law
- b.) Avogadro's law
- c.) Boyle's law
- d.) Charles's law

**Q3.** The mass % of carbon in  $\text{CO}_2$  is \_\_\_\_.

- a.) 0.034%
- b.) 27.27%
- c.) 3.4%
- d.) 28.7%

**Q4.** The empirical formula of benzene is \_\_\_\_.

- a.) CH
- b.)  $\text{C}_6\text{H}_6$
- c.)  $\text{C}_3\text{H}_3$
- d.)  $\text{C}_2\text{H}_2$

**Q5.** What is the mole fraction of the solute in a 1.00 molal aqueous solution?

- a.) 0.00177
- b.) 0.0344
- c.) 0.0177
- d.) 0.1770

**Q6.** State the number of significant figures in each of the following numbers:

- i.) 207.35

# Hrw Chemistry Ch 1

**Sabine Zange**



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