

Potentiometry

Choose the correct answer:

No	Sentence	answer
1	All of the following are related to reference electrode except: a) of known and constant potential b) obeys Nernst equation. c) of high resistance to permit current of electric current. d) insensitive to activity of analyte solution.	C
2	Potential of standard hydrogen electrode is: a) Zero b) related to H^+ concentration c) 0.059 PH d) A and B e) All of the above	d
3	The increase in chloride ion concentration in Ag/AgCl electrode leads to: a) increase the potential b) decrease the potential c) no effect d) none of them	b
4	Potential of calomel electrode is proportional to chloride ion concentration. a) directly b) inversely c) equal d) no relation	b
5	In silver/silver chloride electrode, as KCl concentration increases: a) silver ions increases, potential increases. b) silver ions decreases, potential increases. c) silver ions decreases, potential remains the same. d) silver ions decreases, potential decreases.	d
6	In calomel electrode, as KCl concentration increases: a) mercury(I) ions increases, potential increases. b) mercury(I) ions decreases, potential increases. c) mercury(I) ions decreases, potential remains the same. d) mercury(I) ions decreases, potential decreases.	d
7	What is the electrode potential for a half-cell consisting of a cadmium electrode immersed in a solution that is 0.0150 M in Cd^{2+} ($E^0 = -0.403$ V) a) 0.457 b) 0.403 c) -0.457 d) -0.201	C

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Table of Contents Mcq On Potentiometry

1. Understanding the eBook Mcq On Potentiometry
 - The Rise of Digital Reading Mcq On Potentiometry
 - Advantages of eBooks Over Traditional Books
2. Identifying Mcq On Potentiometry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mcq On Potentiometry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mcq On Potentiometry
 - Personalized Recommendations
 - Mcq On Potentiometry User Reviews and Ratings
 - Mcq On Potentiometry and Bestseller Lists
5. Accessing Mcq On Potentiometry Free and Paid eBooks
 - Mcq On Potentiometry Public Domain eBooks
 - Mcq On Potentiometry eBook Subscription Services
 - Mcq On Potentiometry Budget-Friendly Options

6. Navigating Mcq On Potentiometry eBook Formats
 - ePub, PDF, MOBI, and More
 - Mcq On Potentiometry Compatibility with Devices
 - Mcq On Potentiometry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mcq On Potentiometry
 - Highlighting and Note-Taking Mcq On Potentiometry
 - Interactive Elements Mcq On Potentiometry
8. Staying Engaged with Mcq On Potentiometry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mcq On Potentiometry
9. Balancing eBooks and Physical Books Mcq On Potentiometry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mcq On Potentiometry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Mcq On Potentiometry
 - Setting Reading Goals Mcq On Potentiometry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mcq On Potentiometry
 - Fact-Checking eBook Content of Mcq On Potentiometry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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