

Section A

For each question there are four possible answers **A**, **B**, **C** and **D**. Choose the **one** you consider to be correct.

Use of the Data Booklet may be appropriate for some questions.

- 1 Compound X is an organic compound that contains 30.6% carbon, 3.8% hydrogen, 20.4% oxygen and 45.2% chlorine by mass.

What is the empirical formula of X?

- A** C_2H_3OCl **B** C_2H_4OCl **C** C_3H_4OCl **D** $C_4H_3O_2Cl_2$

- 2 A sample of propane, C_3H_8 , with a mass of 9.61 g is completely combusted in an excess of oxygen under room conditions.

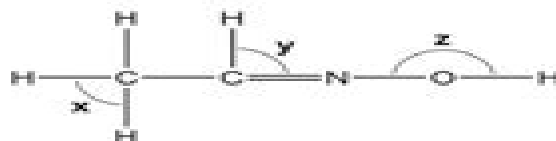
Which volume of carbon dioxide gas is produced?

- A** 4.89 dm^3 **B** 5.24 dm^3 **C** 14.7 dm^3 **D** 15.7 dm^3

- 3 Which atom has the same number of electrons as an ammonium ion?

- A** Mg **B** Na **C** Ne **D** O

- 4 Ethanal reacts with hydroxylamine, NH_2OH , to form the molecule shown.



What is the order of **increasing** bond angle in this structure from smallest to largest?

- A** z, x, y **B** y, z, x **C** x, z, y **D** z, y, x

May 2013 9701 A Level Chemistry Paper

David Baud



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