#### INTERMEDIATE 1STYEAR -CHEMISTRY

#### MODEL QUESTION PAPER ( EFFECTIVE FROM IPE MARCH 2013)

TIME: 3 HOURS]

[ MAX. MARKS : 60

SECTION-A

10x2 = 20

#### NOTE: Answer all the questions

- Calculate the R M S speed of nitrogen molecules at 27°C?
- 2. What is disproportionation reaction? Give example.
- Calculate the pH of 0.05M Ba(OH)<sub>2</sub> solution.
- Write the biological importance of magnesium and calcium.
- Give the reactions that takes place at anode and cathode in Castner-Kellner process.
- 6. Graphite is a good conductor of electricity, Why?
- 7. What is allotropy? Name two crystalline allotropes of carbon.
- 8. Define the terms COD and BOD?
- What are the harmful effects of acid rains?
- G ive the structural formulae of the following compounds
  (a)2,3- dimethyl butane (b) 2-methyl but-1-ene

SECTION-B

6x4 = 24

#### NOTE: Answer any Six of the following questions.

- 11. Write any four postulates of kinetic molecular theory of gases.
- 12. Balance the following redox equation in acidic medium by ion-electron method

MnO<sub>4</sub>- + C<sub>2</sub>O<sub>4</sub><sup>2-</sup> → Mn<sup>2+</sup> + CO<sub>2</sub>

- State Hess's law of constant heat summation. Explain with one example.
- Derive the relation between Kp and Kc for the equilibrium reaction

 $N_{2}(g) + 3H_{2}(g) \rightarrow 2 NH_{3}(g)$ 

- 15. Explain electrolytic method of preparation of hydrogen peroxide.
- 16. Explain the structure of diborane on the basis of VBT
- Explain (a) position Isomerism and (b) functional group Isomerism with one example for each of them.
- Explain (a) Wurtz reaction and (b) Friedel- Craft's alkylation with one example for each of them.

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