

সুতরাং, বিশেষ  $MY_2$  অম্লজিহ্ন হবে।



**↑↑↑↑ ivlv fv+jv:**

কিছু ভাষ্যমাত্রায় কোনো প্রকণে একটি লবণ প্রস্তুত থাকবে যদি অম্লজিহ্ন হবে তা নির্ভর করে লবণটির প্রাকৃতিক তপনতের উপর। যদি প্রকণে কোনো লবণের উপস্থিতি অম্লজিহ্নের আয়নিক তপনতের মান লবণটির প্রাকৃতিক তপনতের মানকে অতিক্রম করতে না পারে তবে লবণটি প্রকণে প্রস্তুত অসম্ভব থাকে। আর যদি লবণের আয়নিক তপনতের মান ( $K_{sp}$ ) লবণটির প্রাকৃতিক তপনতের মান ( $K_{sp}$ ) কে অতিক্রম করে তবে লবণটি প্রকণ থেকে অম্লজিহ্ন হয়।

অর্থাৎ,

$K_{sp} < K_{sp}$  : প্রকণটি অসম্পূর্ণ প্রকণ

$K_{sp} = K_{sp}$  : প্রকণটি সম্পূর্ণ প্রকণ

$K_{sp} > K_{sp}$  : প্রকণ থেকে প্রব অম্লজিহ্ন হয়

III A পাঠের PN প্রকণ তৈরি করলে তা নিম্নলিখিত অম্লজিহ্ন হয়,



সুতরাং, PN প্রকণের ঘনমাত্রা,  $0.1M$  হবে—

$P^+$  আয়নের ঘনমাত্রা,  $[P^+] = 0.1 M$

$N^-$  আয়নের ঘনমাত্রা,  $[N^-]_1 = 0.1 M$

এখানে, PN প্রকণে  $N^-$  আয়ন উপস্থিত। অন্যদিকে A পাঠের  $MIN_3$  লবণের একটি অম্লজিহ্ন  $N^-$  উপস্থিত থাকলে A পাঠের PN প্রকণ তৈরি করলে সম্ভাব্যতঃ প্রতিক্রিয়া করলে  $MIN_3$  এর প্রাকৃতিক তপনত হবে।

যদি,  $MIN_3$  এর লবণ প্রাকৃতিক, S

প্রকণে  $MIN_3$  নিম্নলিখিত অম্লজিহ্ন হয়,



সুতরাং,  $M^{3+}$  আয়নের ঘনমাত্রা,  $[M^{3+}] = S$

$N^-$  আয়নের ঘনমাত্রা,  $[N^-]_2 = 3S$

সেহেতু,  $N^-$  আয়ন  $PN$  ও  $MIN_3$  উভয় ক্ষেত্রে উপস্থিত, সেহেতু

$N^-$  আয়নের মোট ঘনমাত্রা হবে,  $[N^-] = (3S + 0.1)M$

এখন  $MIN_3$  এর আয়নিক তপনত,

$$K_{sp} = [M^{3+}] \times [N^-]^3$$

$$\begin{aligned} \text{কি, } 2.7 \times 10^{-11} &= S \times (3S + 0.1)^3 \quad [\because \text{কি, } K_{sp} = 2.7 \times 10^{-11}] \\ &= S \times [(3S)^3 + 3 \times (3S)^2 \times 0.1 + 3 \times 3S \times (0.1)^2 + (0.1)^3] \\ &= S \times [27S^3 + 2.7S^2 + 0.09S + 0.001] \\ &= 27S^4 + 2.7S^3 + 0.09S^2 + 0.001S \end{aligned}$$

$$\text{কি, } 2.7 \times 10^{-11} = 0.001S \quad [\because S^4, S^3, S^2 \text{ এর মান খুবই নগণ্য}]$$

$$\begin{aligned} \therefore S &= \frac{2.7 \times 10^{-11}}{0.001} \\ &= 2.7 \times 10^{-8} \text{ mol L}^{-1} \end{aligned}$$

অর্থাৎ,  $MIN_3$  ক্ষেত্রে লবণ প্রাকৃতিক,  $2.7 \times 10^{-8} \text{ mol L}^{-1}$

অতএব, A পাঠের PN প্রকণ তৈরি করলে  $MIN_3$  এর প্রাকৃতিক  $10^{-8} \text{ mol L}^{-1}$  থেকে-এম পাঠের  $2.7 \times 10^{-8} \text{ mol L}^{-1}$  হবে।

# Hsc Chemistry Note Bd

**J Spring**



## **Hsc Chemistry Note Bd:**

## Reviewing **Hsc Chemistry Note Bd**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Hsc Chemistry Note Bd**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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