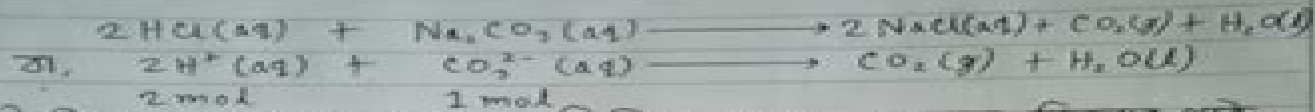


NAME OF THE EXPERIMENT: 0.1M Na_2CO_3	Page No.	
উদ্ব. দ্বারা সংরক্ষিত নমুনা উদ্ব. (HCl/উদ্ব.)	Expt No.	01
চলমান দিন	Date :	

মূলনীতি:

Na_2CO_3 ও HCl এর প্রকৃষ্ট বিক্রিয়া সমীকরণ নিম্নরূপ:



টাইট্রেশনের তৈরিতে, প্রকৃষ্ট বিক্রিয়া কোষ প্রকৃষ্ট বিক্রিয়ায় পরিণত হয়, CO_3^{2-} এর মোল সংখ্যা (ফ্রাক্ট) $\equiv \text{H}^+$ এর মোল সংখ্যা (ক্যালি ব্রোমিডে দেয়া)

প্রয়োজনীয় সামগ্র্যনিক বস্তু:

- 1) 0.1M Na_2CO_3 উদ্ব.
- 2) 0.1M HCl উদ্ব.
- 3) নির্দেশক : মিথাইল অরেঞ্জ

প্রয়োজনীয় যন্ত্রপাতি:

- 1) কনিক্যাল ফ্লাস্ক
- 2) ব্যালিট
- 3) পিপেট
- 4) ক্যাম্পমই মেটাল

কার্যের ধারা:

- 1) প্রস্তুত করা 0.1M HCl উদ্ব. (প্রকৃষ্টাতি অম্ল উদ্ব.) অংশ নমুনা HCl দ্বারা ব্যালিট বর্ণ করে এবং মেটালের সাথে ক্যাম্পমই ফ্লাস্ক খালিভাবে আঁকিয়ে নিই।
- 2) প্রস্তুত করা 0.1M Na_2CO_3 উদ্ব. (প্রকৃষ্ট উদ্ব.) এর আনুমানিক ফ্রাক্ট থেকে পিপেট দ্বারা 10ml উদ্ব. কনিক্যাল ফ্লাস্কে নিই, কনিক্যাল ফ্লাস্কে উদ্ব. 1-2 ফোটা মিথাইল অরেঞ্জ যোগ করে, তখন উদ্ব. বর্ণ হলুদ হয়।
- 3) কনিক্যাল ফ্লাস্কে তৈরি একটি সাদা সোজা টুকরা রাখি, ব্যালিট HCl অম্ল উদ্ব. মটি নিয়ে ঘোঁটা করে 'গম পাঠ' - এ ফেরত করে, চিহ্নমতে রাস মাতে ব্যালিটের বর্ণ লক্ষ্য করে ব্যালিটের HCl উদ্ব. ঘোঁটা ঘোঁটা কনিক্যাল ফ্লাস্কে যোগ করে

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