HSC Maharashtra Board question paper: March 2013

Note:

- i. All question are compulsory
- Answer to the two sections are to be written in the same answer book.
- iii. Figure to the right hand side indicate full marks.
- Write balanced chemical equations and draw neat and labelled diagrams wherever necessary.
- v. Encery new question must be started on a new page.
- vi. Use of logarithmic table is allowed

CHEMISTRY: SECTION - I

Q. 1. Select and write the most appropriate answer from the given alternatives for each sub-question: [7]

- i... In body centred cubic structure the space occupied is about (A) 68.96 (C)38 % (ID) 32 96 16. For a gaseous reaction the unit of rate of reaction is atm mol s-1 L atm s-1 (III) atm s 1 CO (ID) most s
- iii. Which of the following compounds contains S = O as well as S = S bonds?
 (A) Sulphuric acid
 (B) Thiosulphuric acid
 (C) Sulphurous acid
 (D) Thiosulphurous acid
- iv. Which of the following solutions shows maximum depression in freezing point?
 (A) 0.5 M Li₂SO₄
 (B) 1 M NaCl
 (C) 0.5 M Al₂(SO₄)₃
 (D) 0.5 M BaCl₂
- V. For a chemical reaction ΔS = 0.035 kJ/K and ΔH = 20kJ.
 (A) 5.14 K
 (B) 57.14 K
 (C) 571.4 K
 (D) 5714.0 K
- vi. The standard e.m.f of the following cell is 0.463 V $Cu \mid Cu_{(DO)}^{**} \mid Ag_{(DO)}^{*} \mid Ag. \text{ If } E_{A_{0}}^{O} = 0.800 \text{ V},$ What is the standard potential of Cu electrode?

 (A) 1.137 V(B) 0.337 V(C) 0.463 V(D) -0.463 V
- vii. Fe₂O₃ is reduced to spongy iron near the top of blast furance by
 (A) H₂ (B) CaO

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M. Cusiter, V. Cusiter

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