Chemistry Final Exam Review Questions

Useful Values and Equations

Some of the following values and equations may be useful for answering some of the questions and problems.

9P = 1.8(9C) + 32speed of light = 3.00×10^8 m/s $c = \lambda w$

R = 0.082057 L-atm / deg-mole Planck's constant = 6.63 × 10-34 J·s Fig. 1800

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- One of the following does not describe gases. Which one is it?
 - a. Expands infinitely.

Easily compressed.

Rigid shape.

- d. Relatively low densities.
- The following properties describe the element, mercury (Hg). Which one is a chemical property?
 - a. Its density is 5.43 g/cm³.
 - b. It is a silvery-white metal.

- c. It is a liquid at 298 K.
- d. Mercury(II) sulfide forms when it is exposed to sulfide ions.
- All of the following are properties of SO₂. Which one is a physical property?
 - It reacts with oxygen to form SO₃².

- e. It is a strong reducing agent.
- mospheric pressure.
- It condenses to a colorless liquid at 14°F and at It reacts with water and oxygen to form an acid...
- 4. Which of the following numbers has 4 significant figures?

b. 0.0430

c. 0.0431

- 0.43980
- 5. How many significant figures should there be in the answer to the following math equation? 2.834 + 5.71520 + 2.12 + 178.1 + 250.2619
 - a. 3

e. 5

- 6. How many significant figures should there be in the answer to the following math equation? $23.413 \text{ g} + (2.15 \text{ cm} \times 1.1 \text{ cm} \times 3.73 \text{ cm})$

b. 3

- 7. At 25°C, one cm3 of iron has a mass of 7.62 grams. What volume of iron is occupied by 125 grams of iron?
 - a. 0.82 cm³

- b. 8.2 cm³
- c. 0.164 cm³

- d. 16.4 cm³
- 8. A cube of lead has a volume of 6.61 cm3. Its mass is 75.0 grams. What is its density?
 - a. 3.85 g/cm³
- b. 7.41 g/cm³
- c. 11.3 g/cm³
- d. 22.6 g/cm³

Modeling Chemistry Final Exam Review Answers

Daniel Bloch

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is known about learning and teaching and provide professional development and collegiality from elementary teachers to college professors As a pedagogy it has been shown to be effective in a variety of content areas and at different educational levels This is an introduction to the process and the community Every POGIL classroom is different and is a reflection of the uniqueness of the particular context the institution department physical space student body and instructor but follows a common structure in which students work cooperatively in self managed small groups of three or four The group work is focused on activities that are carefully designed and scaffolded to enable students to develop important concepts or to deepen and refine their understanding of those ideas or concepts for themselves based entirely on data provided in class not on prior reading of the textbook or other introduction to the topic The learning environment is structured to support the development of process skills such as teamwork effective communication information processing problem solving and critical thinking The instructor's role is to facilitate the development of student concepts and process skills not to simply deliver content to the students The first part of this book introduces the theoretical and philosophical foundations of POGIL pedagogy and summarizes the literature demonstrating its efficacy. The second part of the book focusses on implementing POGIL covering the formation and effective management of student teams offering guidance on the selection and writing of POGIL activities as well as on facilitation teaching large classes and assessment The book concludes with examples of implementation in STEM and non STEM disciplines as well as guidance on how to get started Appendices provide additional resources and information about The POGIL Project <u>Digital Learning and Teaching in Chemistry</u> Yehudit Dori, Courtney Ngai, Gabriela Szteinberg, 2023-07-12 Education is always evolving and most recently has shifted to increased online or remote learning Digital Learning and Teaching in Chemistry compiles the established and emerging trends in this field specifically within the context of learning and teaching in chemistry This book shares insights about five major themes best practices for teaching and learning digitally digital learning platforms virtual visualisation and laboratory to promote learning in science digital assessment and building communities of learners and educators The authors are chemistry instructors and researchers from nine countries contributing an international perspective on digital learning and teaching in chemistry While the chapters in this book span a wide variety of topics as a whole they focus on using technology and digital platforms as a method for supporting inclusive and meaningful learning The best practices and recommendations shared by the authors are highly relevant for modern chemistry education as teaching and learning through digital methods is likely to persist Furthermore teaching chemistry digitally has the potential to bring greater equity to the field of chemistry education in terms of who has access to quality learning and this book will contribute to that goal This book will be essential reading for those working in chemical education and teaching Yehudit Judy Dori is internationally recognised formerly Dean of the Faculty of Education of Science and Technology at the Technion Israel Institute of Technology and won the 2020 NARST Distinguished Contributions to Science Education through Research Award DCRA for her exceptional research contributions

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Selected Water Resources Abstracts, 1989

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