

## CHAPTER 7 PRACTICE TEST: ATOMIC STRUCTURE &amp; PERIODIC TRENDS- KEY

1) When a solution of sodium chloride is vaporized in a flame, the color of the flame is:

- A) Blue      **B) Yellow**      C) Green      D) Violet      E) White

2) Electromagnetic radiation travels through vacuum at a speed of \_\_\_\_\_ m/s.

- A) 186,000      **C)  $3.00 \times 10^8$**       E) It depends on  $\lambda$   
B) 125      D) 10,000

3) The wavelength of light that has a frequency of  $1.20 \times 10^{13} \text{ s}^{-1}$  is \_\_\_\_\_ m.

- A) 25.0      **B)  $2.50 \times 10^{-5}$**       C) 0.0400      E) 2.5  
D) 12.0

GIVEN:  $\nu = 1.20 \times 10^{13} \text{ s}^{-1}$

UNKNOWN:  $\lambda$

CONNECTION:  $c = \lambda \nu$

SUBSTITUTE & SOLVE:  $\lambda = c/\nu$

$$\lambda = \frac{3.00 \times 10^8 \text{ m} \cdot \text{s}^{-1}}{1.20 \times 10^{13} \text{ s}^{-1}}$$

$$\lambda = 2.50 \times 10^{-5} \text{ m}$$

4) Ham radio operators often broadcast on the 6-meter band. The frequency of this EM radiation is \_\_\_\_\_ MHz.

- A) 500      B) 200      **C) 50**      D) 20      E) 2.0

GIVEN:  $\lambda = 6 \text{ m}$

UNKNOWN:  $\nu$  (Must convert answer to MHz!)

CONNECTION:  $c = \lambda \nu$

SUBSTITUTE & SOLVE:  $\nu = c/\lambda$

$$\nu = \frac{3.00 \times 10^8 \text{ m} \cdot \text{s}^{-1}}{6 \text{ m}}$$

$$\nu = 0.5 \times 10^8 \text{ Hz} \times \frac{1 \text{ MHz}}{10^6 \text{ Hz}} = 50 \text{ MHz}$$

5) The energy of a photon of light is \_\_\_\_\_ proportional to its frequency and \_\_\_\_\_ proportional to its wavelength.

- A) directly, directly      C) inversely, directly      E) indirectly, not  
B) inversely, inversely      **D) directly, inversely**

6) The energy of a photon that has a wavelength of 12.3 nm is \_\_\_\_\_ J.

- A)  $1.51 \times 10^{-17}$       B)  $4.42 \times 10^{-23}$       C)  $1.99 \times 10^{-23}$       D)  $2.72 \times 10^{-10}$       **E)  $1.62 \times 10^{-17}$**

7) The wavelength of a photon that has an energy of  $5.25 \times 10^{-19} \text{ J}$  is \_\_\_\_\_ m.

- A)  $3.79 \times 10^{-7}$**       B)  $2.64 \times 10^6$       C)  $2.38 \times 10^{23}$       D)  $4.21 \times 10^{-24}$       E)  $3.79 \times 10^7$

8) A mole of red photons of wavelength 725 nm has \_\_\_\_\_ kJ of energy.

- A)  $2.74 \times 10^{-19}$       B)  $4.56 \times 10^{-40}$       C)  $6.05 \times 10^{-7}$       **D) 165**      E) 227

GIVENS:  $\lambda = 725 \text{ nm}$  (must convert to m before using formulas)

$$\lambda = 725 \text{ nm} \times \frac{1 \text{ m}}{10^9 \text{ nm}} = 7.25 \times 10^{-7} \text{ m}$$

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**Didier Musso**



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Jespersen, Pamela Kerrigan, 2023-07-04 Always study with the most up to date prep Look for AP Chemistry Premium 2025 Prep Book with 6 Practice Tests Comprehensive Review Online Practice ISBN 9781506291802 on sale July 2 2024 Publisher's Note Products purchased from third party sellers are not guaranteed by the publisher for quality authenticity or access to any online entities included with the product

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Courtney A. Young, 2019-02-01 This landmark publication distills the body of knowledge that characterizes mineral processing and extractive metallurgy as disciplinary fields It will inspire and inform current and future generations of minerals and metallurgy professionals Mineral processing and extractive metallurgy are atypical disciplines requiring a combination of knowledge experience and art Investing in this trove of valuable information is a must for all those involved in the industry students engineers mill managers and operators More than 192 internationally recognized experts have contributed to the handbook's 128 thought provoking chapters that examine nearly every aspect of mineral processing and extractive metallurgy This inclusive reference addresses the magnitude of traditional industry topics and also addresses the new technologies and important cultural and social issues that are important today Contents Mineral Characterization and Analysis Management and Reporting Comminution Classification and Washing Transport and Storage Physical Separations Flotation Solid and Liquid Separation Disposal Hydrometallurgy Pyrometallurgy Processing of Selected Metals Minerals and Materials

### **Resources in Education**, 1999

Graduate Programs in the Physical Sciences, Mathematics, Agricultural Sciences, the Environment & Natural Resources 2011 (Grad 4) Peterson's, 2011-05-01 Peterson's Graduate

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### **Selected**

**Water Resources Abstracts**, 1988-04

**Heath Science Connections 9** Jack L. Candido, Edward S. James, Ronald E.

Phillips, 1987

**Progress in Modeling and Simulation of Batteries** John Turner, 2016-06-15 Modeling and simulation of

batteries in conjunction with theory and experiment are important research tools that offer opportunities for advancement of technologies that are critical to electric motors The development of data from the application of these tools can provide the basis for managerial and technical decision making Together these will continue to transform batteries for electric vehicles This collection of nine papers presents the modeling and simulation of batteries and the continuing contribution being made to this impressive progress including topics that cover Thermal behavior and characteristics Battery management system design and analysis Moderately high fidelity 3D capabilities Optimization Techniques and Durability As electric vehicles continue to gain interest from manufacturers and consumers alike improvements in economy and affordability as well as adoption of alternative fuel sources to meet government mandates are driving battery research and development Progress in modeling and simulation will continue to contribute to battery improvements that deliver increased power energy storage and durability to further enhance the appeal of electric vehicles

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