

Chemistry I – Unit 5 Review

13. Balance each:

- iron
- acetic acid
- hydrogen's reaction
- Mass % composition
- empirical formula
- molecular formula

14. Find the molar mass of the following:

- SCN_2 _____
- $\text{Fe}(\text{NO}_3)_3$ _____
- Ag_2SO_4 _____
- $\text{C}_2\text{H}_6\text{O}$ _____
- $\text{C}_2\text{H}_4\text{Cl}_2$ _____
- $\text{C}_2\text{H}_5\text{OH}$ _____

15. Calculate the masses of various isotopes of an element.

Isotope	Mass (g)	Relative mass
Hydrogen	1.008	
Helium	4.003	
Lithium	7.016	
Boron	10.811	

16. Calculate the mass of an element from its isotopes.

17. Explain the difference between mass and weight and the units for each.

Modeling Chemistry Unit 5 Review V

Chao Zhang



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