

## Model 1: Rules of Solubility in Aqueous Solutions

Several solid compounds are placed in water to determine if they are soluble. The results are shown in the chart below. If it does dissolve no mark is made.

The top row shows the cation of the compound. The far left column shows the anion of the compound. For example,  $\text{Mg}(\text{OH})_2$  is insoluble,  $\text{MgBr}_2$  is soluble and  $\text{AgCl}$  is insoluble.

**Table 1** – This table presents an overview of solubility of selected salts in water

**X** = does not dissolve in water.    \* partially soluble.

	$\text{NH}_4^+$	$\text{Li}^+$	$\text{Na}^+$	$\text{K}^+$	$\text{Mg}^{2+}$	$\text{Ca}^{2+}$	$\text{Sr}^{2+}$	$\text{Ba}^{2+}$	$\text{Ag}^+$	$\text{Pb}^{2+}$	$\text{Hg}_2^{2+}$	$\text{Fe}^{3+}$	$\text{Cu}^{2+}$	$\text{Zn}^{2+}$
$\text{NO}_3^-$														
$\text{C}_2\text{H}_3\text{O}_2^-$														
$\text{CO}_3^{2-}$														
$\text{C}_2\text{O}_4^{2-}$														
$\text{Cl}^-$									X	X	X			
$\text{Br}^-$									X	X	X			
$\text{I}^-$									X	X	X			
$\text{SO}_4^{2-}$						X	X	X	X	X	X			
$\text{OH}^-$					X	*	*		X	X	X	X	X	X
$\text{S}^{2-}$					X	X	X		X	X	X	X	X	X
$\text{CO}_3^{2-}$					X	X	X	X	X	X	X	X	X	X
$\text{PO}_4^{3-}$					X	X	X	X	X	X	X	X	X	X
$\text{C}_2\text{O}_4^{2-}$					X	X	X	X	X	X	X	X	X	X
$\text{CrO}_4^{2-}$					X	X	X	X	X	X	X	X	X	X

### Key Questions

1. Is calcium carbonate soluble or insoluble?
2. Is silver bromide soluble or insoluble?
3. Is iron (III) sulfate soluble or insoluble?
4. For what cations are the compounds always soluble in water?
5. For what anions are the compounds always soluble in water?
6. For what anions are most of the compounds usually soluble (with some exceptions)?
7. For those anions that usually form soluble compounds, which cations result in the

# Model Rules Of Solubility In Aqueous Solutions Answers

**EW Minium**



## **Model Rules Of Solubility In Aqueous Solutions Answers:**

The book delves into Model Rules Of Solubility In Aqueous Solutions Answers. Model Rules Of Solubility In Aqueous Solutions Answers is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Model Rules Of Solubility In Aqueous Solutions Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
    - Chapter 1: Introduction to Model Rules Of Solubility In Aqueous Solutions Answers
    - Chapter 2: Essential Elements of Model Rules Of Solubility In Aqueous Solutions Answers
    - Chapter 3: Model Rules Of Solubility In Aqueous Solutions Answers in Everyday Life
    - Chapter 4: Model Rules Of Solubility In Aqueous Solutions Answers in Specific Contexts
    - Chapter 5: Conclusion
  2. In chapter 1, the author will provide an overview of Model Rules Of Solubility In Aqueous Solutions Answers. The first chapter will explore what Model Rules Of Solubility In Aqueous Solutions Answers is, why Model Rules Of Solubility In Aqueous Solutions Answers is vital, and how to effectively learn about Model Rules Of Solubility In Aqueous Solutions Answers.
  3. In chapter 2, the author will delve into the foundational concepts of Model Rules Of Solubility In Aqueous Solutions Answers. The second chapter will elucidate the essential principles that need to be understood to grasp Model Rules Of Solubility In Aqueous Solutions Answers in its entirety.
  4. In chapter 3, this book will examine the practical applications of Model Rules Of Solubility In Aqueous Solutions Answers in daily life. This chapter will showcase real-world examples of how Model Rules Of Solubility In Aqueous Solutions Answers can be effectively utilized in everyday scenarios.
  5. In chapter 4, this book will scrutinize the relevance of Model Rules Of Solubility In Aqueous Solutions Answers in specific contexts. The fourth chapter will explore how Model Rules Of Solubility In Aqueous Solutions Answers is applied in specialized fields, such as education, business, and technology.
  6. In chapter 5, the author will draw a conclusion about Model Rules Of Solubility In Aqueous Solutions Answers. The final chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Model Rules Of Solubility In Aqueous Solutions Answers.

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