

## PROCEDIMIENTOS DE SERVICIO (Continuación)

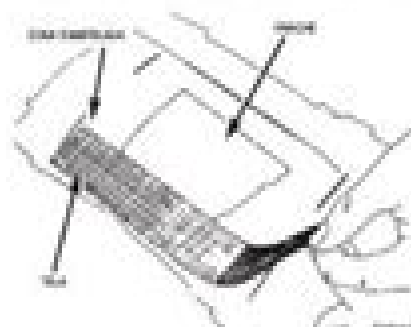


Fig. 13 Recubrimiento de espacio con tela

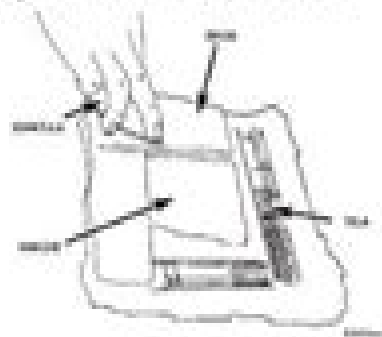


Fig. 14 Recubrimiento de tela con papel

## DESMONTAJE E INSTALACION

## TRAVESANO DE FALSO BASTIDOR DE SOPORTE DEL RADADOR

## DESMONTAJE

(1) Desenganche el pestillo del capó y abra el capó.  
(2) Retire los suplementos de piezas que fijan la placa protectora rejilla al travesaño de falso bastidor de soporte del radiador.

(3) Retire las pernos que sujetan los apoyos a la parte inferior del travesaño de falso bastidor de soporte del radiador.

(4) Retire las pernos que sujetan el travesaño de falso bastidor a la planchita de acero del radiador.

(5) Retire las tuercas que sujetan el pestillo del capó al travesaño de falso bastidor de soporte del radiador.

(6) Separe del vehículo el travesaño de falso bastidor de soporte del radiador.

## INSTALACION

Instalar el procedimiento anterior.

## REJILLA - CIRROS

## DESMONTAJE

(1) Retire la placa protectora del parabombas delantero. Para informarse sobre los procedimientos apropiados, consulte el grupo T3, Parabombas y bombas.

(2) Con una llave adecuada, retire los tornillos que sujetan la rejilla a la placa protectora delantera.

(3) Separe la rejilla de la placa protectora (Fig. 15).



Fig. 15 Rejilla

## INSTALACION

Instalar el procedimiento anterior.

## PESTILLO DEL CAPO

## DESMONTAJE

(1) Desenganche el pestillo del capó y abra el capó.  
(2) Marque el contorno del pestillo del capó en el travesaño de falso bastidor de soporte del radiador para facilitar la instalación.

(3) Retire las tuercas que sujetan el pestillo del capó al travesaño de falso bastidor de soporte del radiador (Fig. 16).

(4) Separe el pestillo del capó del travesaño de falso bastidor.

(5) Desenganche de la articulación de la caja del pestillo la cavadura del cable de desenganche del capó.

(6) Desenganche el extremo del cable de la palanca de desenganche del pestillo del capó.

(7) Separe el pestillo del capó del vehículo.

## INSTALACION

(1) Coloque el pestillo del capó en el vehículo.

## INFORMACION GENERAL (Continuación)

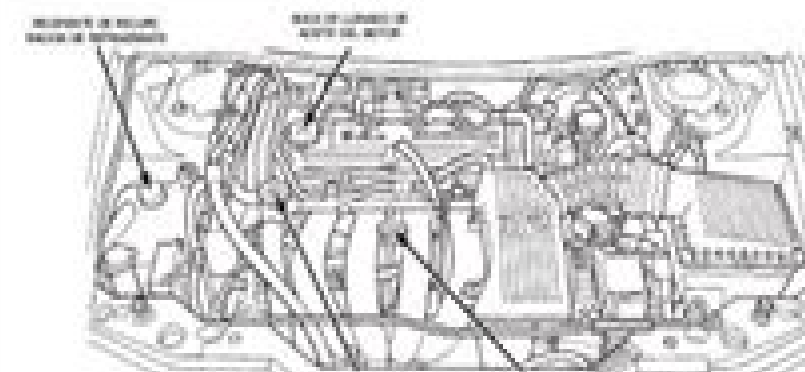


Fig. 7 Localización de la varilla indicadora y boca de llenado de aceite del motor-2.4L

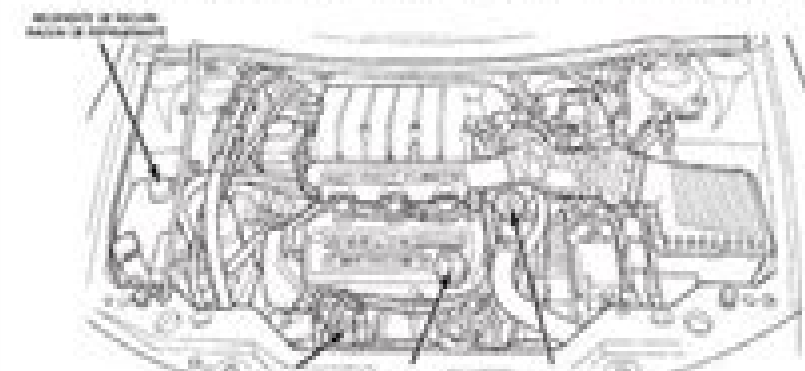


Fig. 8 Localización de la varilla indicadora y la boca de llenado de aceite del motor-2.4L



Fig. 9 Nivel de aceite

## VISCOSIDAD SAE

El grado de viscosidad SAE se utiliza para especificar la viscosidad del aceite del motor. Utilice solo aceites de motor de viscosidades múltiples, tales como 10W-30 o 10W-40. Entre aceites se especifican con un grado de viscosidad SAE doble que cubren la gama de viscosidades para temperaturas frías a calientes. Seleccione el aceite de motor que mejor se

adapte al tiempo concreto de temperatura y velocidad (Fig. 10).

NIVEL DE VISCOSIDAD DEL ACEITE DEL MOTOR												
10W-30 Preferred												
10W-40												
°C	-30°	-20°	-10°	0°	10°	20°	30°	40°	50°	60°	70°	80°
°F	-20°	-4°	10°	32°	50°	68°	86°	104°	122°	140°	158°	176°
Según la temperatura ambiente antes del arranque motor de arranque												

Fig. 10 Temperatura y viscosidad del aceite del motor

# Manual De Stratus

**Jiyuan Zhang**



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web  $m \times c \times \Delta t$  equation 1 where  $m$  is the total mass of the solution solute plus solvent  $c$  is the specific heat of the solution and  $\Delta t$  is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of *guided inquiry designing the ideal hand warmer* - Dec 27 2022

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web background the hand warmers are filled with sodium acetate click the metal disc and you will notice the liquid changes state and becomes a solid in the process heat is released the sodium acetate is a supersaturated solution

*hand warmer chemistry easy chemical hot packs science* - Apr 30 2023

web oct 9 2021 sodium acetate coin or other small piece of metal plastic bags seal the coin inside a small bag place this bag inside a larger bag containing supersaturated sodium acetate get this by dissolving as much sodium acetate as possible in hot water this chemical hand warmer is reusable

*ap chem hand warmer chemistry lab including answers* - Sep 04 2023

web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

**hand warmer challenge random chem information studocu** - Apr 18 2022

web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution  $\Delta H_{\text{soln}}$  or the amount of energy released or absorbed in kJ/mol of the substance being dissolved

ap chemistry designing an effective hand warmer - Jan 28 2023

web mar 4 2014 contains water the hand warmer is struck in a manner that ruptures the inner pouch releasing the ionic salt into the water of the outer pouch the salt dissolves and the water warms this type of hand warmer tends to produce a more vigorous heat than the dry powder type of hand warmer but does not produce heat for quite as long

**designing a hand warmer lab youtube** - Mar 30 2023

web feb 2 2017 this is a video outlining a flinn lab on designing a hand warmer it was created by alex brinley charis conwell and siena joy for our ap chemistry class

**explore the chemistry within hand warmers science project** - Sep 23 2022

web hand warmers provide a unique and fun way to study the chemistry of crystal formation and heat generation by pressing a button in a pouch which contains a supercooled solution you start a rapid exothermic heat producing crystallization in this science fair project you will determine how the starting temperature affects hand warmer chemistry

**ap chemistry hand warmer lab youtube** - Oct 25 2022

web feb 1 2017 hannah boster seth lewis and william makinen s recorded powerpoint presentation summarizing our work on the flinn handwarmer lab created for mr kerns ap

12 the hand warmer design challenge studocu - Nov 25 2022

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**ap chemistry lab 10 hand warmer design challenge big idea 5** - Oct 05 2023

web lab investigation in today s experiment you will design and execute an experimental procedure to determine which of three ionic compounds is most suitable for use in a hand warmer the three compounds you will use is determined by the group number you are assigned the ideal hand warmer