

Key

Ideal Gas Law Worksheet $PV = nRT$

Use the ideal gas law, "PerV-nRT", and the universal gas constant $R = 0.0821 \frac{L \cdot atm}{K \cdot mol}$ to solve the following problems:

If pressure is needed in kPa then convert by multiplying by $101.3 kPa / 1 atm$ to get
 $R = 8.31 kPa \cdot L / (K \cdot mole)$

- 1) If I have 4 moles of a gas at a pressure of 5.6 atm and a volume of 12 liters, what is the temperature?

$$PV = nRT$$

$$T = \frac{PV}{nR} = \frac{(5.6 atm)(12 L)}{4 mol \cdot 0.0821 \frac{L \cdot atm}{K \cdot mol}}$$

$$T = 204.63 K$$

- 2) If I have an unknown quantity of gas at a pressure of 1.2 atm, a volume of 31 liters, and a temperature of 87 °C, how many moles of gas do I have?

$$PV = nRT$$

$$n = \frac{PV}{RT} = \frac{1.2 atm \cdot 31 L}{0.0821 \frac{L \cdot atm}{K \cdot mol} \cdot 360 K}$$

$$n = 1.2586 mol$$

- 3) If I contain 3 moles of gas in a container with a volume of 60 liters and at a temperature of 400 K, what is the pressure inside the container?

$$PV = nRT$$

$$P = \frac{nRT}{V} = \frac{3 mol \cdot 0.0821 \frac{L \cdot atm}{K \cdot mol} \cdot 400 K}{60 L}$$

$$P = 1.642 atm$$

or $P = 166.29 kPa$

- 4) If I have 7.7 moles of gas at a pressure of 0.09 atm and at a temperature of 56 °C, what is the volume of the container that the gas is in?

$$PV = nRT$$

$$V = \frac{nRT}{P} = \frac{7.7 mol \cdot 0.0821 \frac{L \cdot atm}{K \cdot mol} \cdot 327 K}{0.09 atm}$$

$$V = 2310.93 L$$

- 5) If I have 17 moles of gas at a temperature of 67 °C, and a volume of 88.89 liters, what is the pressure of the gas?

$$PV = nRT$$

$$P = \frac{nRT}{V} = \frac{17 mol \cdot 0.0821 \frac{L \cdot atm}{K \cdot mol} \cdot 390 K}{88.89 L}$$

$$P = 5.34 atm$$

or $P = 540.61 kPa$

- 6) If I have an unknown quantity of gas at a pressure of 0.5 atm, a volume of 25 liters, and a temperature of 300 K, how many moles of gas do I have?

$$PV = nRT$$

$$n = \frac{PV}{RT} = \frac{0.5 atm \cdot 25 L}{0.0821 \frac{L \cdot atm}{K \cdot mol} \cdot 300 K}$$

$$n = 0.5075 mol$$

Ideal Gas Law Problems Sheet 2 Answers

**Frank Schaffer Publications, Joan
DiStasio**



Ideal Gas Law Problems Sheet 2 Answers:

A Guide to Teaching in the Active Learning Classroom Paul Baepler, J. D. Walker, D. Christopher Brooks, Kem Saichaie, Christina I. Petersen, 2023-07-03 While Active Learning Classrooms or ALCs offer rich new environments for learning they present many new challenges to faculty because among other things they eliminate the room's central focal point and disrupt the conventional seating plan to which faculty and students have become accustomed. The importance of learning how to use these classrooms well and to capitalize on their special features is paramount. The potential they represent can be realized only when they facilitate improved learning outcomes and engage students in the learning process in a manner different from traditional classrooms and lecture halls. This book provides an introduction to ALCs briefly covering their history and then synthesizing the research on these spaces to provide faculty with empirically based practical guidance on how to use these unfamiliar spaces effectively. Among the questions this book addresses are: How can instructors mitigate the apparent lack of a central focal point in the space? What types of learning activities work well in the ALCs and take advantage of the affordances of the room? How can teachers address familiar classroom management challenges in these unfamiliar spaces? If assessment and rapid feedback are critical in active learning, how do they work in a room filled with circular tables and no central focus point? How do instructors balance group learning with the needs of the larger class? How can students be held accountable when many will necessarily have their backs facing the instructor? How can instructors evaluate the effectiveness of their teaching in these spaces? This book is intended for faculty preparing to teach in or already working in this new classroom environment for administrators planning to create ALCs or experimenting with provisionally designed rooms and for faculty developers helping teachers transition to using these new spaces. **Merrill Chemistry**

Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Unique Scientific Puzzles** Dr. S. Pancharatnam, 2020-04-06 Born and brought up in a sugar factory village Pancharatnam turned into a good scholar in leading school and college in Pune then IIT Bombay. This propelled him into some of the world's best universities: UC Berkeley and Stanford. He managed to get away from the ivory towers of USA and devote to more challenging and rewarding Indian chemical industry with more useful R&D and project engineering. Another success story was his own business of specialty filters for the mech engineering industry with over hundred reputed customers. So here he is with a fully enjoyable career of fifty years with over fifty projects and many publications. Having spent all his life in technical investigation and improvements he has brought to you vast variety of 500 interesting puzzles from various fields. Most are actually encountered in daily life. Many are truly unique and some quite advanced. Further over 500 jokes are added for relaxing in between. So go ahead, struggle, laugh and learn a lot. This small book is highly recommended for students of final years of school, all college students in science, engineering and tech professionals. Even teachers will find it interesting for setting tests. Of course riddles and easy puzzles can be enjoyed by everyone. Free quarterly updates are assured on your email id for 1 year. **Knowing Thermodynamics** Nicole Marie

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The Ideal Gas Law Handbook - Everything You Need to Know about Ideal Gas Law Patrick Hurley,2016-04-29 This book is your ultimate Ideal gas law resource Here you will find the most up to date information facts quotes and much more In easy to read chapters with extensive references and links to get you to know all there is to know about Ideal gas law s whole picture right away Get countless Ideal gas law facts right at your fingertips with this essential resource The Ideal gas law Handbook is the single and largest Ideal gas law reference book This compendium of information is the authoritative source for all your entertainment reference and learning needs It will be your go to source for any Ideal gas law questions A mind tickling encyclopedia on Ideal gas law a treat in its entirety and an oasis of learning about what you don t yet know but are glad you found The Ideal gas law Handbook will answer all of your needs and much more *Ideal Gases* Lifeliqe,2019 This lesson plan covers the ideal gas law and the different values for the ideal gas constant how to make various calculations using the ideal gas law and explains the conditions under which real gases are most or least ideal

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Table of Contents Ideal Gas Law Problems Sheet 2 Answers

1. Understanding the eBook Ideal Gas Law Problems Sheet 2 Answers
 - The Rise of Digital Reading Ideal Gas Law Problems Sheet 2 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ideal Gas Law Problems Sheet 2 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ideal Gas Law Problems Sheet 2 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ideal Gas Law Problems Sheet 2 Answers
 - Personalized Recommendations
 - Ideal Gas Law Problems Sheet 2 Answers User Reviews and Ratings
 - Ideal Gas Law Problems Sheet 2 Answers and Bestseller Lists
5. Accessing Ideal Gas Law Problems Sheet 2 Answers Free and Paid eBooks
 - Ideal Gas Law Problems Sheet 2 Answers Public Domain eBooks
 - Ideal Gas Law Problems Sheet 2 Answers eBook Subscription Services
 - Ideal Gas Law Problems Sheet 2 Answers Budget-Friendly Options

6. Navigating Ideal Gas Law Problems Sheet 2 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ideal Gas Law Problems Sheet 2 Answers Compatibility with Devices
 - Ideal Gas Law Problems Sheet 2 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ideal Gas Law Problems Sheet 2 Answers
 - Highlighting and Note-Taking Ideal Gas Law Problems Sheet 2 Answers
 - Interactive Elements Ideal Gas Law Problems Sheet 2 Answers
8. Staying Engaged with Ideal Gas Law Problems Sheet 2 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ideal Gas Law Problems Sheet 2 Answers
9. Balancing eBooks and Physical Books Ideal Gas Law Problems Sheet 2 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ideal Gas Law Problems Sheet 2 Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ideal Gas Law Problems Sheet 2 Answers
 - Setting Reading Goals Ideal Gas Law Problems Sheet 2 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ideal Gas Law Problems Sheet 2 Answers
 - Fact-Checking eBook Content of Ideal Gas Law Problems Sheet 2 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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