

21222

**3 Hours / 70 Marks**

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15 minutes extra for each hour

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*Instructions – (1) All Questions are Compulsory.*

*(2) Answer each next main Question on a new page.*

*(3) Illustrate your answers with neat sketches wherever necessary.*

*(4) Figures to the right indicate full marks.*

*(5) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.*

**Marks**

**1. Attempt any FIVE of the following.**

**10**

a) State two advantages and two disadvantages of oil hydraulic systems.

# Hydraulics And Pneumatics Question Paper

**Lauren Gardner**



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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid  $\text{H}_2\text{CO}_3$  and its conjugate base the bicarbonate ion  $\text{HCO}_3^-$

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a binary acid is an acid that consists of hydrogen and one other element the most common binary acids contain a halogen the

acid name begins with the prefix hydro followed by the base name of the anion followed by the suffix ic

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oct 18 2023 most of the acids are solid it has a sour taste they produce hydrogen after reacting with active metals like zinc magnesium aluminium or iron acids have a low ph value i e below 7 salts and water are produced when an acid reacts with base or alkalis most of the acids are solids

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arrhenius s definition of acids and bases the earliest definition of acids and bases is arrhenius s definition which states that an acid is a substance that forms hydrogen ions h when dissolved in water and a base is a substance that forms hydroxide ions oh when dissolved in water

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other s chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is ce acid base water salt nonnumber

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