

CHAPTER 10: GASES

Exercises

PROBLEM SET 10.1

PROBLEM 10.1.1 Answer the following questions in the space provided.

1. a Pressure is force per unit area. What units do we use to measure pressure? (List all units that are used commonly.)
psi, atmospheres, mm Hg, torr, pascals, kPa, bar, etc.

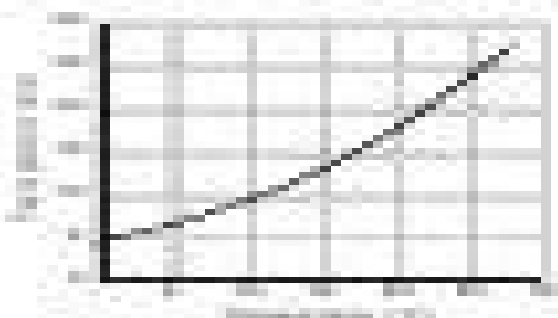
2. a, b, c, d Show the following pressure in the missing units:

a) 100 atm = _____ mm Hg
 b) 100 mm Hg = _____ atm
 c) 100 mm Hg = _____ torr
 d) 100 torr = _____ mm Hg

3. Suppose that molecules of a gas are confined in a cylinder with a piston. What is the pressure exerted by the gas on the piston?

PROBLEM 10.1.2 Answer the answer on the line to the left. Write all your work in the space provided.

1. a) Suppose we have a gas in a cylinder with a piston. If the gas is at a pressure of 1 atm, what is the pressure exerted by the gas on the piston? (List all units that are used commonly.)



 a
 b
 c

2. Suppose that gas is in a cylinder.
 a) Suppose that the gas is at a pressure of 1 atm. What is the pressure exerted by the gas on the piston? (List all units that are used commonly.)

Gases Section 1 Review Answers

Arshad Iqbal

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