

Wintersemester 2019/2020 - Klausuraufgaben 8B

Aufgabe 8 (max. 70 Punkte/100% Gesamtpunkte)

Die nachstehenden Probleme betreffen die Veranschaulichung des Verhaltens der Stoffeigenschaften mittels "Diagramme" von Molanteile, Druck, Temperatur in der Zustandsform. Bitte führen Sie alle Schritte der Berechnungen und die Skizzen an!

$$R = 8,314 \text{ J mol}^{-1} \text{ K}^{-1} = 0,08314 \text{ bar dm}^3 \text{ mol}^{-1} \text{ K}^{-1}$$

1. Chlorwasserstoff (Molare Masse $M = 36,5 \text{ g mol}^{-1}$) wird bei einer Temperatur von 25°C in einem geschlossenen Zylinder bei einem Druck von 10 bar (absolut) gehalten.

a) Wie groß ist die molare Dichte ρ_m in mol dm^{-3} ?
 $\rho_m = 1,23 \text{ mol dm}^{-3}$ (2,46 g dm^{-3})

2. Chlorwasserstoff wird bei einer Temperatur von 25°C bei 10 bar (absolut) in einem geschlossenen Zylinder bei einem Druck von 10 bar (absolut) gehalten.

b) Wie groß ist die molare Dichte ρ_m in mol dm^{-3} ?
 $\rho_m = 1,23 \text{ mol dm}^{-3}$ (2,46 g dm^{-3})

3. Chlorwasserstoff wird bei einer Temperatur von 25°C bei 10 bar (absolut) in einem geschlossenen Zylinder bei einem Druck von 10 bar (absolut) gehalten. Die Temperatur wird auf 100°C erhöht. Wie groß ist die molare Dichte ρ_m in mol dm^{-3} ?

c) Wie groß ist die molare Dichte ρ_m in mol dm^{-3} ?
 $\rho_m = 1,23 \text{ mol dm}^{-3}$ (2,46 g dm^{-3})

4. In einem geschlossenen Zylinder wird bei einer Temperatur von 25°C bei 10 bar (absolut) in einem geschlossenen Zylinder bei einem Druck von 10 bar (absolut) gehalten. Die Temperatur wird auf 100°C erhöht. Wie groß ist die molare Dichte ρ_m in mol dm^{-3} ?

d) Wie groß ist die molare Dichte ρ_m in mol dm^{-3} ?
 $\rho_m = 1,23 \text{ mol dm}^{-3}$ (2,46 g dm^{-3})

5. Chlorwasserstoff wird bei einer Temperatur von 25°C bei 10 bar (absolut) in einem geschlossenen Zylinder bei einem Druck von 10 bar (absolut) gehalten. Die Temperatur wird auf 100°C erhöht. Wie groß ist die molare Dichte ρ_m in mol dm^{-3} ?

e) Wie groß ist die molare Dichte ρ_m in mol dm^{-3} ?
 $\rho_m = 1,23 \text{ mol dm}^{-3}$ (2,46 g dm^{-3})

6. Chlorwasserstoff wird bei einer Temperatur von 25°C bei 10 bar (absolut) in einem geschlossenen Zylinder bei einem Druck von 10 bar (absolut) gehalten. Die Temperatur wird auf 100°C erhöht. Wie groß ist die molare Dichte ρ_m in mol dm^{-3} ?

Gases Section Answers

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Content 1 Solid State 2 Solutions 3 Electro Chemistry 4 Chemical Kinetics 5 Surface Chemistry 6 General Principles And Processes Of Isolation Of Elements 7 P Block Elements 8 D And F Block Elements 9 Coordination Compounds And Organometallics 10 Haloalkanes And Haloarenes 11 Alcohols Phenols And Ethers 12 Aldehydes Ketones And Carboxylic Acids 13 Organic Compounds Containing Nitrogen 14 Biomolecules 15 Polymers 16 Chemistry In Everyday Life Appendix 1 Important Name Reactions And Process 2 Some Important Organic Conversion 3 Some Important Distinctions

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