

Mixed Gas Law Worksheet

Answers available at end

Name: _____ (show your work) period: _____

Convert:

$$100 \text{ mmHg} = \text{_____ kPa}$$

$$300 \text{ mmHg} = \text{_____ atm}$$

$$120 \text{ atm} = \text{_____ mmHg}$$

$$300 \text{ torr} = \text{_____ kPa}$$

$$25 \text{ kPa} = \text{_____ mmHg}$$

$$25 \text{ mmHg} = \text{_____ atm}$$

$$300^{\circ}\text{C} = \text{_____ K}$$

$$150\text{K} = \text{_____}^{\circ}\text{C}$$

$$5000^{\circ}\text{C} = \text{_____ K}$$

$$300\text{K} = \text{_____}^{\circ}\text{C}$$

$$-255^{\circ}\text{C} = \text{_____ K}$$

$$3500\text{K} = \text{_____}^{\circ}\text{C}$$

$$200^{\circ}\text{C} = \text{_____}^{\circ}\text{F}$$

$$95^{\circ}\text{F} = \text{_____}^{\circ}\text{C} = \text{_____ K}$$

$$45^{\circ}\text{F} = \text{_____}^{\circ}\text{C} = \text{_____ K}$$

- 1) A 70. liter sample of gas initially at 280°C is allowed to cool at constant pressure, what will the new volume be at 0°C ?

$$P_1 = \text{_____} \quad P_2 = \text{_____}$$

$$V_1 = \text{_____} \quad V_2 = \text{_____}$$

$$T_1 = \text{_____} \quad T_2 = \text{_____}$$

- 2) A 4.0 liter sample of gas initially at 25°C is allowed to cool at constant pressure, what will the new volume be at -5°C ?

$$P_1 = \text{_____} \quad P_2 = \text{_____}$$

$$V_1 = \text{_____} \quad V_2 = \text{_____}$$

$$T_1 = \text{_____} \quad T_2 = \text{_____}$$

- 3) A buoyancy vest (used in scuba diving) is filled to a volume of 2.3 liters at a pressure of 1 atmosphere (at the surface where the temperature is a balmy 300K), what will the volume of air be in the vest when the diver dives to a depth of about 90 feet where the temperature is 17°C and the pressure is 4 atmospheres?

$$P_1 = \text{_____} \quad P_2 = \text{_____}$$

Gas Laws And Scuba Diving Chemistry Worksheet

Laurence Whisler



Gas Laws And Scuba Diving Chemistry Worksheet:

Unit 4: Gas Laws, Student Workbook Mike Stieff, 2012-02-14 The Connected Chemistry Curriculum *Introductory Chemistry, Module Five : the Gas Laws* Hoewe, Barbara, Jalil, Shahid, 1976 **Expanding on the Gas Laws**, 2009-01-01 Inquiries in Science Chemistry Series Expanding on the Gas Laws Teacher's Guide **Gas Laws**, 2010 **GAS LAWS** NARAYAN CHANGDER, 2024-04-01 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel <https://www.youtube.com/smartquizziz> I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging quiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today's academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests quizzes and trivia challenges one requires test taking techniques and skills in addition to subject knowledge It also provides you with the skills and information you need to achieve a good score in challenging tests or competitive examinations Whether you have studied the subject on your own read for pleasure or completed coursework it will assess your knowledge and prepare you for competitive exams quizzes trivia and more

Gases and Gas Laws Hobsons Publishing, PLC, 1983-01-01 **Simple Demonstrations of the Gas Laws** William Maurice Dehn, 1908 **Chemistry Jeopardy** Brian Pressley, 2002 A quiz game designed to reinforce key chemistry concepts ideas and facts *Lesson 45* ADvTECH., **The Gas Laws** Laurence Whisler, 1964 *Instant Lessons in Chemistry* Denise De Vreeze, Kath McMicking, 1998 This publication consists of reproducible worksheets usually two pages suitable for senior high school Chemistry The worksheets aim to give students experience in applying concepts interpreting and presenting data and building a core of chemistry knowledge *Chemistry Lesson Plans, Study Guides, and Lecture Notes* Doctor Chemistry, 2017-08-14 The Ultimate Guide to Learning or Teaching Chemistry This book contains the real lecture notes and slide of a highly effective high school and college Chemistry teacher Teachers Never plan another lesson again Students Ace your upcoming exam This series covers all of the topics of High School Chemistry and General Chemistry including Accuracy and Significant Figures Mixtures Metric System Bonding Atomic Theory Periodic Table VSEPR Ionic and Covalent Bonding Geometric Bonding The Mole and Molar Mass Equation Balancing Thermodynamics Stoichiometry States of Matter Gas Laws and Calculations Reaction Calculations Acids and Bases Limiting Reagents Redox and Electro Chemistry Organic Chemistry Basics

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