

Lab Data

Report pressure of water (mmHg) 21.1

Verify your calculation. In your set-up, do your units cancel?

Did you report your data to the correct number of significant figures?

Volume of hydrogen gas collected (mL) 15.00

Volume of hydrogen gas collected (L) .01500

Pressure of hydrogen gas (mmHg) 759.9

Pressure of hydrogen gas (atm)

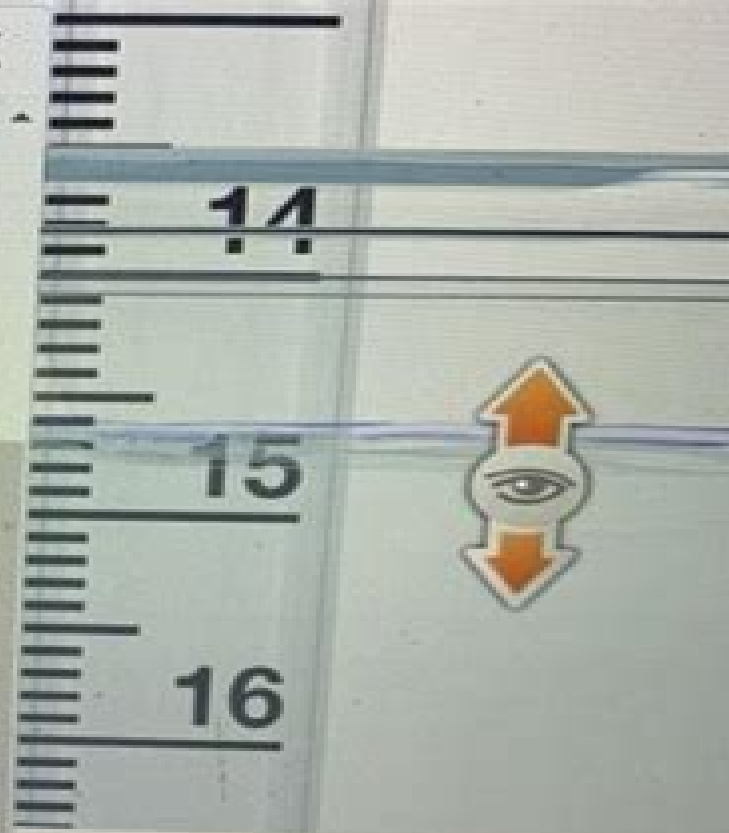
Moles of hydrogen gas produced (mol)

Ideal gas constant ($\text{L}\cdot\text{atm}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$)

How to calculate ideal gas constant

Water Vapor Pressure Table

X



MY NOTES

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Sin A $\frac{a}{c} = \sin B \frac{b}{c}$
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Gas Law Constant Chemistry Lab Report Answers

Carl A. Rouse



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Fetzer-Gislason, Julie Ellefson Kuehn, 2004-02-20 With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment The scenarios are drawn from the lives of people who work with chemistry every day ranging from field ecologists to chemical engineers and include many health professionals as well Exploring General Chemistry in the

Laboratory Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science

Progress in High Temperature Physics and Chemistry Carl A. Rouse, 2013-09-03 Progress in High Temperature Physics and Chemistry Volume 1 is a collection of various theoretical experimental and technological aspects of high temperature phenomena The volume presents broad scientific disciplines such as equilibrium and nonequilibrium thermodynamics ionization and excitation phenomena radiation and other transport phenomena structure and properties of matter sources and measurement or determination of high temperatures chemical physics gas dynamics shock wave phenomena in solids liquids and gases producing high temperatures with associated physical and chemical changes exploding wires and exploding foils peaceful applications of nuclear energy and astrophysics Thermophysics and thermochemistry are briefly described as well Physicists chemists worker and researchers in the field of high temperature physics and chemistry will find the book very useful **Scientific and Technical Aerospace Reports**, 1991

Everything You Need to Ace Chemistry in One Big Fat Notebook Workman Publishing, Jennifer Swanson, 2020-09-22 Chemistry No problem This Big Fat Notebook covers everything you need to know during a year of high school chemistry class breaking down one big bad subject into accessible units Learn to study better and get better grades using mnemonic devices definitions diagrams educational doodles and quizzes to recap it all Including Atoms elements compounds and mixtures The periodic table Quantum theory Bonding The mole Chemical reactions and calculations Gas laws Solubility pH scale Titrations Le Chatelier's principle and much more General, Organic, and Biological Chemistry Study Guide and

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