Gas Laws and SCUBA Diving

Read the accompanying article "Gas Laws & SCUBA Diving," Chambiasters, February 1983, pp. 4-6.
Answer the following questions completely.

- 1. Who does diving 30to below sea level affect our budies more than being in a trailling 30to above sea level?
- 2. What parts of a diver's body are recet affected by pressure cleanges?
- 2. State Boyle's Law.
- 4. Why don't SCURA diver's large collapse as they descend?
- What would happen to a diver who does not axhale while surfacing from a 50 m dive? Explain in terms of Boyle's Law.
- 6. State Harry's Law.
- 7. What gos is associated with causing bubbles in the Mood and other body fluids?
- 8. What is another name for deposit pression sightness?
- 9. Describe how increased pressure in the chamber refueves symptoms of decompression vickness.
- (th. What is nitrogen nancosis?)
- Which gas law explains why air connections of the trace presents of CO) are more dangerous when the total air presents is higher? Explain.
- 12. Is the relationship between the temperature of water and the solubility of a gas in it a direct or inverse relationship? Explain.
- 12. Use your answer to 01.1 to explain why it is dangerous for a diver to take a hot shower after a deep dive.

Gas Laws And Scuba Diving Answer Key

Chao Zhang

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