

## Part A

Charles's law states that the volume ( $V$ ) of a fixed quantity of gas is directly proportional to its temperature ( $T$ ) at a constant pressure.

$$V \propto T$$

You can verify this law by plotting the graph of a gas's volume versus its temperature.

To perform this analysis, pump the handle only once so that a fixed number of gas molecules enter the gas chamber. Set the number of "Heavy Species" gas molecules to 100 using the text box given in the tab named **Gas in chamber**. Once the pressure reaches the value of about 0.50 atm, click on the "Pressure" button under the tab **Constant Parameter**, which is at the top right corner of the simulation. Go to the panel named "Tools and Options." Select the ruler by checking off the option in the Measurement Tools. Observe that the height of the cylinder (as measured left to right) does not remain constant because the molecules exert pressure on the walls of the cylinder.

Set the temperature by using the heat control box to add or remove heat as given in the table below.

Temperature (K)	200.	250.	300.	350.
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Use the ruler to find the height of the cylinder as an average between two whole nanometer markings on the ruler. For example, if the value is fluctuating between 6.0 and 7.0 nm, consider 6.5 nm to be the height of the cylinder even if the ruler appears to hover closer to either marking.

Complete the table below with your raw data for the height of the cylinder at each temperature.

Drag the appropriate labels to their respective targets.

3.5

4.5

5.5

6.5

7.5

8.5

Temperature (K)	200.	250.	300.	350.
Height of cylinder (nm)	4.5	5.5	6.5	7.5

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# Gas Laws Activity Lab Answers Key

**CL Gary**



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**Concepts, Strategies and Models to Enhance Physics Teaching and Learning** Eilish McLoughlin, Paul van Kampen, 2019-07-24 This book discusses novel research on and practices in the field of physics teaching and learning It gathers selected high quality studies that were presented at the GIREP ICPE EPEC 2017 conference which was jointly organised by the International Research Group on Physics Teaching GIREP European Physical Society Physics Education Division and the Physics Education Commission of the International Union of Pure and Applied Physics IUPAP The respective chapters address a wide variety of topics and approaches pursued in various contexts and settings all of which represent valuable contributions to the field of physics education research Examples include the design of curricula and strategies to develop student competencies including knowledge skills attitudes and values workshop approaches to teacher education and pedagogical strategies used to engage and motivate students This book shares essential insights into current research on physics education and will be of interest to physics teachers teacher educators and physics education researchers around the world who are working to combine research and practice in physics teaching and learning

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## Reviewing **Gas Laws Activity Lab Answers Key**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Gas Laws Activity Lab Answers Key**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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