#### temperature:

- a measure of the <u>average kinetic</u> energy of the particles in a sample of matter
- does not depend on the amount of \_\_matter\_\_\_\_\_\_ in the sample
- symbol is <u>⊤</u> : unit is <u>c</u>

#### heat:

- total amount of kinetic energy that flows because of a difference in tempature.
- depends on mass \_\_\_\_\_ of sample

Kinetic energy is energy in motion

Potential energy is store

Potential energy is hiding and cannot be

measured

Only <u>changes</u> in P.E. can be measured.

#### specific heat capacity:

- symbol is \_\_\_\_\_\_\_; unit is \_\_\_\_\_\_\_\_



When heat (Q) is absorbed by a system, part of it (C) goes into storage as \_\_\_\_\_ energy and part of it is used to make the molecules move around \_\_\_\_\_, raising the \_\_\_\_\_ (\Delta t).

Why does sand get hotter in the day and colder at night than the water?

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