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First Semester B.Tech Degree Regular and Supplementary Examination December 2020 (2019 scheme)

Course Code: CYT100 Course Name: ENGINEERING CHEMISTRY (2019-Scheme)

	(2017-Centure)	
Max. N	Marks: 100 Duration:	3 Hours
	PART A Answer all questions, each carries 3 marks.	Marks
1	What will be the standard electrode potential of Ni2* / Ni electrode if the cell	(3)
	potential of the cell Ni / Ni ^{2*} (1M) // Cu ² (0.1M) / Cu is 0.59 V at 25 °C?	
	$E^0_{Cu2+vCu} = 0.34 \text{ V}$	
2	Briefly explain the principle of electroless plating.	(3)
3	Give the mechanism of interaction of electromagnetic radiation with oscillating	(3)
	dipole.	
4	State Beer-Lambert's law and write the differential form.	(3)
5	Write any three differences of TGA and DTA.	(3)
6	Explain sol-gel method for the synthesis of nano particles	(3)
7	Determine the configuration of the following alkenes as E or Z:	(3)
	H ₃ C CH ₂ OH C ₆ H ₅ COOH A B	
8	Mention any three advantages of OLEDs over LED and LCD.	(3)
9	Explain disinfection by chlorination.	(3)
10	Compare BOD and COD.	(3)
	PART B Answer one full question from each module, each question carries 14 marks	
11 a)	•	(10)
	conditions.	
ъ)	How is the cell constant of a conductivity cell measured?	(4)

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Joacim Rocklöv

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