

2.5

ELEMENTS AND COMPOUNDS

Section Review

Objectives

1. Explain the difference between elements and a compound
2. Distinguish between a substance and a mixture
3. Identify the chemical symbols for elements and write a chemical formula for a compound

Vocabulary

- | | |
|--------------|----------------|
| 1. element | 2. compound |
| 3. substance | 4. homogeneous |

Part A Completion

Fill in the blank with the word or phrase that best completes or explains each statement. Write your answer in the space provided. (10 points)

- | | |
|--|---|
| 1. An element is a substance that cannot be broken down into simpler substances by chemical means. | 6. A compound is a substance that is made up of two or more different elements. |
| 2. A compound is a substance that is made up of two or more different elements. | 7. A mixture is a combination of two or more substances that are not chemically combined. |
| 3. A substance is a material that has a definite composition and properties. | 8. A homogeneous mixture is a mixture in which the components are uniformly distributed. |
| 4. A mixture is a combination of two or more substances that are not chemically combined. | 9. A heterogeneous mixture is a mixture in which the components are not uniformly distributed. |
| 5. A chemical formula is a symbolic representation of a chemical compound. | 10. A chemical reaction is a process in which one or more substances are converted into one or more different substances. |

Elements And Compounds Section Review

Guillaume Favre

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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