

### Activity A:

#### Subatomic particles

#### Get the Gizmo ready:

- Use the arrows to create an atom with two protons, two neutrons, and two electrons.
- Turn on **Show element name**.



**Question:** What are the properties of protons, neutrons, and electrons?

1. **Observe:** Turn on **Show element symbol** and **Element notation**. Three numbers surround the element symbol: the **mass number** ( $A$ ), electrical charge (no number is displayed if the atom is neutral), and the **atomic number** ( $Z$ ).
2. **Investigate:** Watch how the numbers change as you add or remove particles.



- A. Which number is equal to the number of protons in the atom?
- B. How can you calculate the number of neutrons ( $N$ ) in an atom?
- C. Which particle (proton, neutron, or electron) has a  
Positive charge?  
Negative charge?

neutrons

by the number of protons

electron

proton

Electrical charge?

neutrons

3. **Observe:** Turn on **Electron shells**. Electrons are arranged in layers called shells. The number of shells is equal to the atomic number ( $Z$ ). The number of electrons in each shell is also equal to the atomic number ( $Z$ ). The number of protons in each shell is also equal to the atomic number ( $Z$ ). The number of neutrons in each shell is also equal to the atomic number ( $Z$ ).

- A. What are the shells called?
- B. What are the shells called?
- C. What are the shells called?

Electron shells

Electron shells

Electron shells

4. **Observe:** Turn on **Electron shells**. The following questions are asked:

- A. How many electrons are in an atom of Helium?
- B. How many electrons are in an atom of Hydrogen?
- C. What is the mass number of an atom with 2 protons and 2 neutrons?
- D. What is the mass number of an atom with 2 protons and 2 neutrons?
- E. How many electrons are in Helium?

2

1

4

4

2

# Element Builder Gizmo Exploration Sheet Answer Key

**Kathleen Armour**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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