

# Electron Configuration

Element	Total electrons	Orbital diagram					Electron configuration	
		1s	2s	2p			3s	
Li	3	<div>↑↓</div>	<div>↑</div>	<div></div>	<div></div>	<div></div>	<div></div>	$1s^2 2s^1$
Be	4	<div>↑↓</div>	<div>↑↓</div>	<div></div>	<div></div>	<div></div>	<div></div>	$1s^2 2s^2$
B	5	<div>↑↓</div>	<div>↑↓</div>	<div>↑</div>	<div></div>	<div></div>	<div></div>	$1s^2 2s^2 2p^1$
C	6	<div>↑↓</div>	<div>↑↓</div>	<div>↑</div>	<div>↑</div>	<div></div>	<div></div>	$1s^2 2s^2 2p^2$
N	7	<div>↑↓</div>	<div>↑↓</div>	<div>↑</div>	<div>↑</div>	<div>↑</div>	<div></div>	$1s^2 2s^2 2p^3$
Ne	10	<div>↑↓</div>	<div>↑↓</div>	<div>↑↓</div>	<div>↑↓</div>	<div>↑↓</div>	<div></div>	$1s^2 2s^2 2p^6$
Na	11	<div>↑↓</div>	<div>↑↓</div>	<div>↑↓</div>	<div>↑↓</div>	<div>↑↓</div>	<div>↑</div>	$1s^2 2s^2 2p^6 3s^1$

# Electron Configuration Review

**Christopher See, Chris John Tyreman**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the authors' names.

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